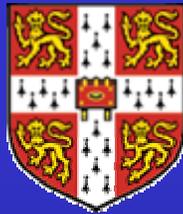


Materials for nuclear waste immobilization:
The effect of point defects in zircon

Miguel Pruneda



*Department of Earth Sciences
University of Cambridge*

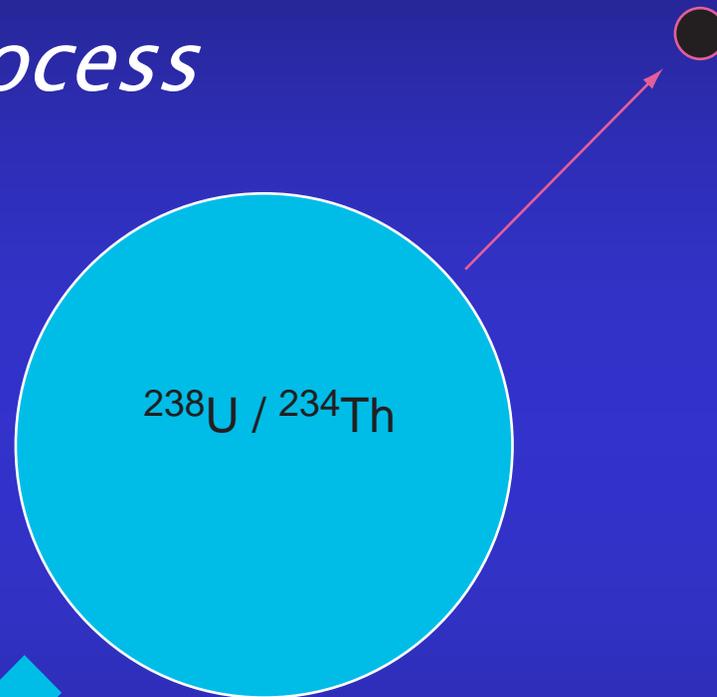


*Cambridge Centre for
Ceramic Immobilisation*



Radiation damage

α -decay process



α -particle

$\sim 5 \text{ MeV}$

It causes:

- Amorphisation
(metamict)*
- Swelling*
- Cracks*
- Leaching*

Recoil



$\sim 100 \text{ keV}$

Zircon: model study: old natural samples

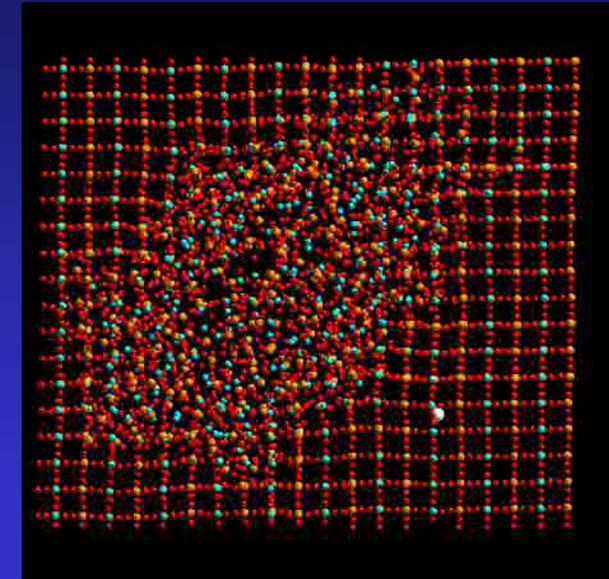
Why first principles?

- Atomistic description

MD simulations

- Complex structures

Empirical potentials



- Complex chemistry

Zr, Ti, Sn, Ca, La, Gd,...

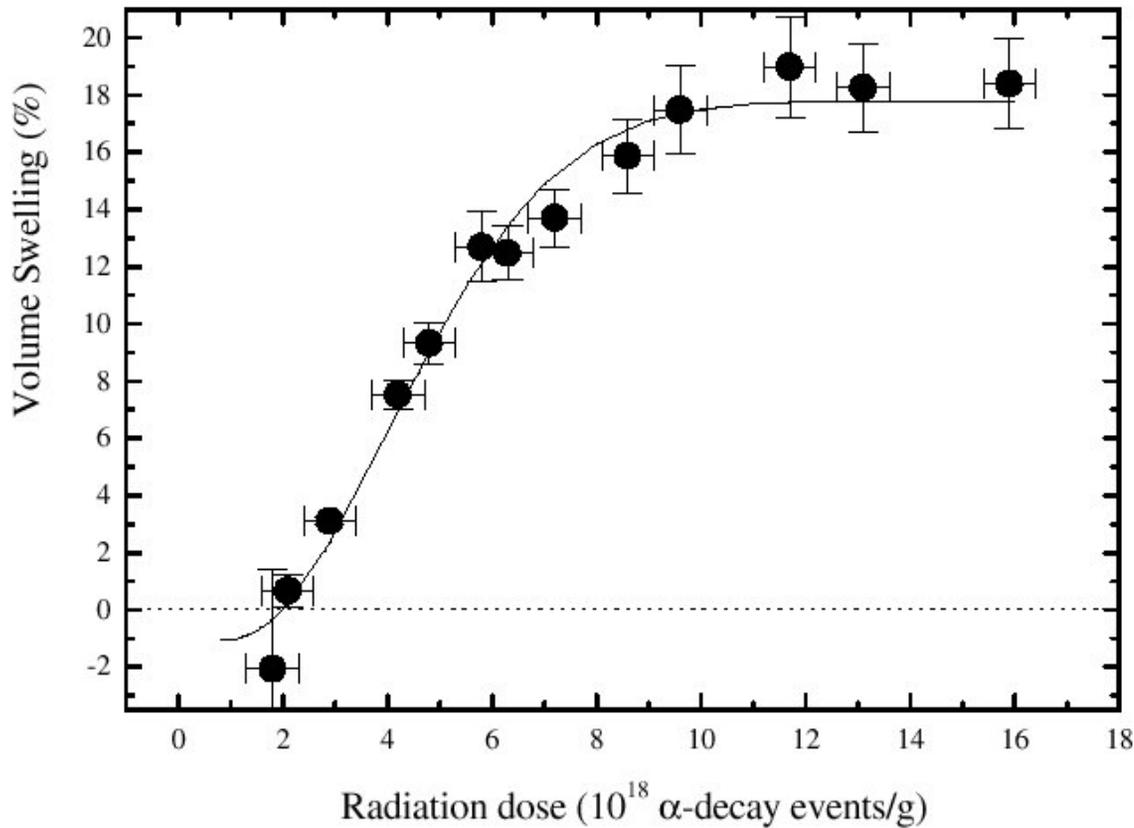
atomic number		atomic weight		symbol		name		state		preparation		isotope	
14	28.09	Si	Silicon	black	solid	synthetically prepared	most stable isotope						

alkali metals	alkaline earth metals	transitional metals	other metals	nonmetals	noble gases
H, Li, Na, K, Rb, Cs, Fr	Be, Mg, Ca, Sr, Ba, Ra	Sc, Ti, V, Cr, Mn, Fe, Co, Ni, Cu, Zn, Ga, Ge, As, Se, Br, Kr	Y, Zr, Nb, Mo, Tc, Ru, Rh, Pd, Ag, Cd, In, Sn, Sb, Te, I, Xe	B, C, N, O, P, S, Cl, Ar	He, Ne, Ar, Kr, Xe, Rn

Lanthanide series																											
58	140.12	59	140.91	60	144.24	61	(145)	62	150.36	63	151.96	64	157.25	65	158.93	66	162.50	67	164.93	68	167.26	69	168.93	70	173.04	71	174.99
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu														
Cerium	Praseodymium	Neodymium	Promethium	Samarium	Europium	Gadolinium	Terbium	Dysprosium	Holmium	Erbium	Thulium	Ytterbium	Lutetium														

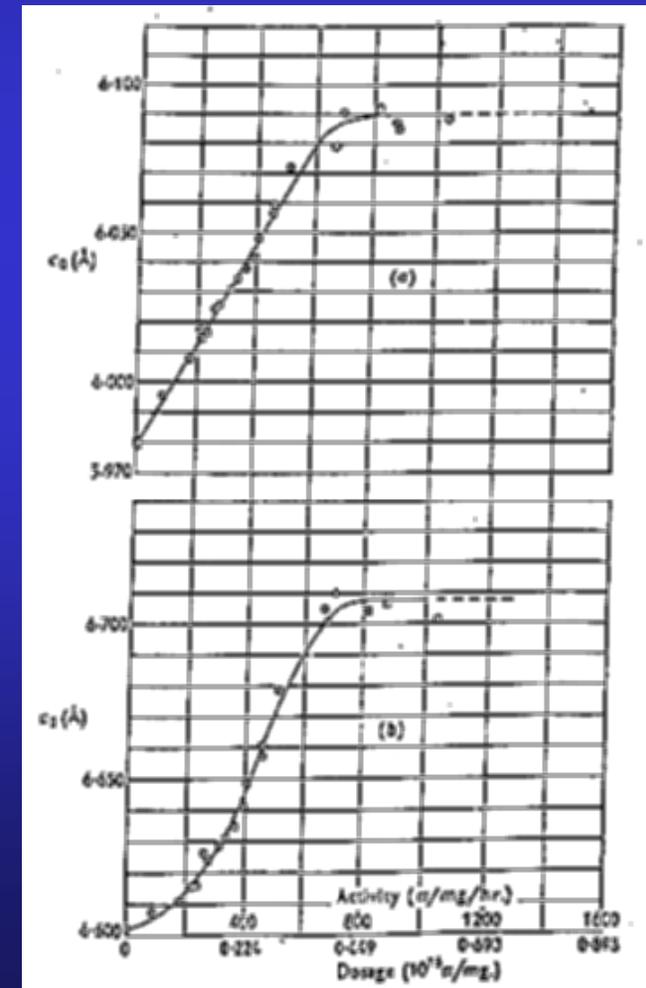
Actinide series																											
90	232.04	91	231.04	92	238.03	93	237.05	94	(244)	95	(243)	96	(247)	97	(247)	98	(251)	99	(252)	100	(257)	101	(258)	102	(259)	103	(262)
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr														
Thorium	Protactinium	Uranium	Neptunium	Plutonium	Americium	Curium	Berkelium	Californium	Einsteinium	Fermium	Mendelevium	Nobelium	Lawrencium														

Swelling in zircon ($ZrSiO_4$)



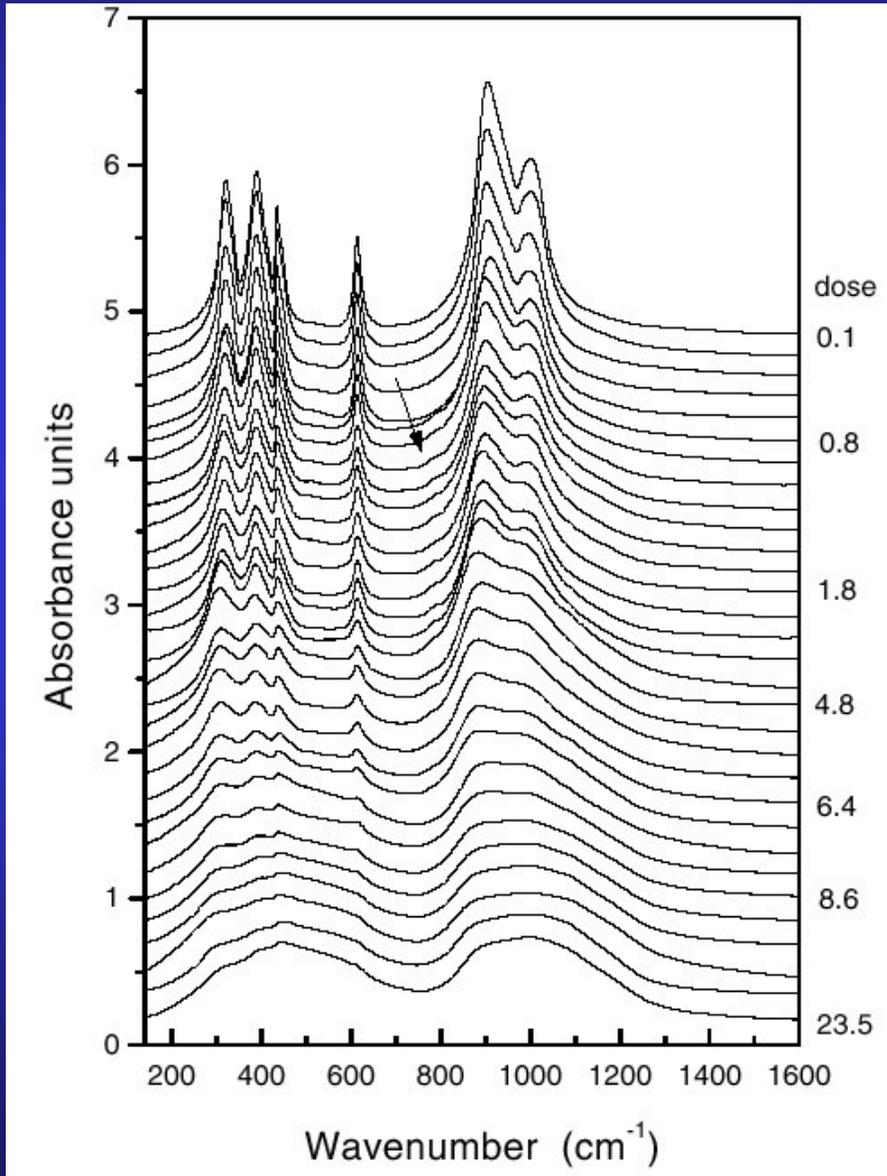
X-Rays...

Crystalline swelling:
lattice parameters vs dose



Total: ~20%
Crystalline: ~5% anisotropic

IR, Raman and NMR spectroscopies



IR

NMR

I. Farnan & E.K.H Salje, JAP **89**, 2084 (2001)

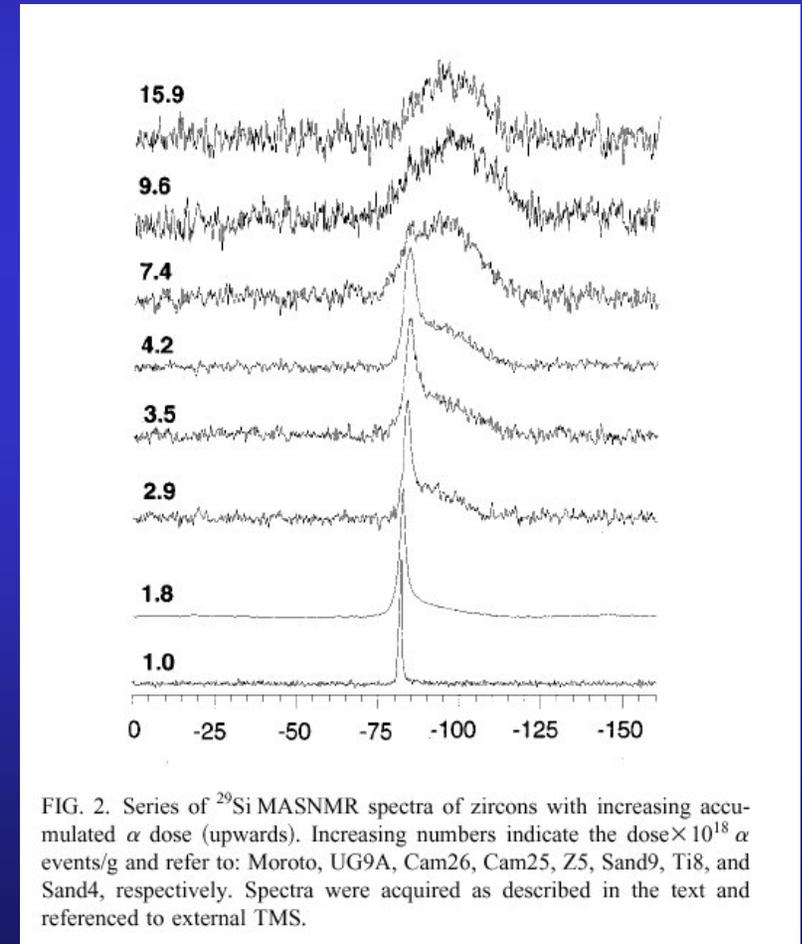


FIG. 2. Series of ^{29}Si MASNMR spectra of zircons with increasing accumulated α dose (upwards). Increasing numbers indicate the dose $\times 10^{18}$ α events/g and refer to: Moroto, UG9A, Cam26, Cam25, Z5, Sand9, Ti8, and Sand4, respectively. Spectra were acquired as described in the text and referenced to external TMS.

M

Localized defects in radiation-damaged zircon (Rios et. al. *Acta Cryst.* (2000) B56, 947)

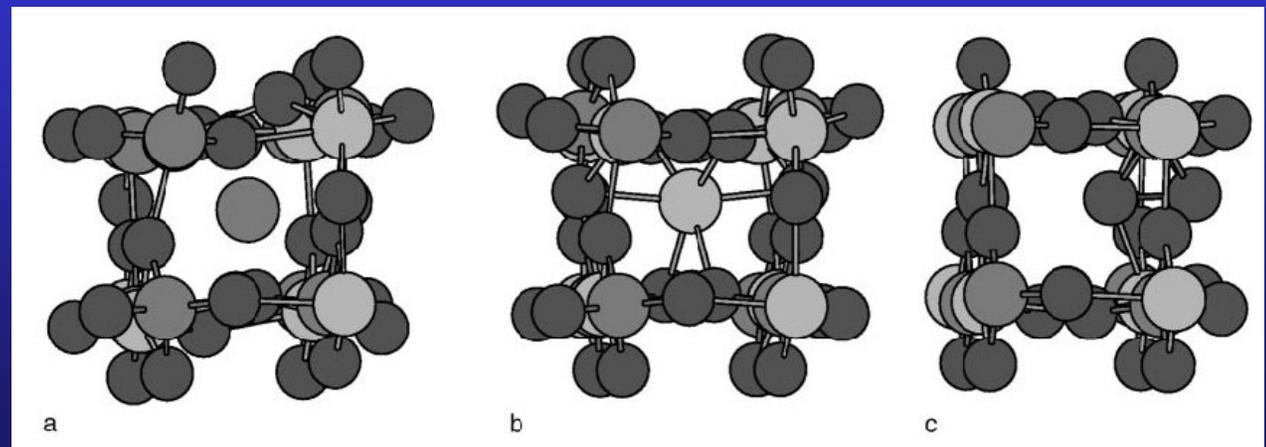
X-ray diffraction experiments in natural samples. (1.8×10^{18} α -decay/g)

- SiO₄ tetrahedra remain essentially undistorted
- Larger anisotropic displacement parameters found for Zr and O atoms

DFT calculations:

Formation energies for a choice of chemical potentials

High values of the formation energies for all defects, except oxygen interstitial



Our method

Linear-scaling DFT based on NAOs (Numerical Atomic Orbitals)



P. Ordejon, E. Artacho & J. M. Soler , Phys. Rev. B 53, R10441
(1996)

- *Born–Oppenheimer* (relaxations, mol. dynamics)
- *DFT* (LDA, GGA)
- *Pseudopotentials* (norm conserving, factorised)
- *Numerical atomic orbitals as basis* (finite range)
- *Numerical evaluation of matrix elements* (3D grid)

Implemented in the SIESTA program

D. Sanchez–Portal, P. Ordejon, E. Artacho & J. M. Soler
Int. J. Quantum Chem. 65, 453 (1997)

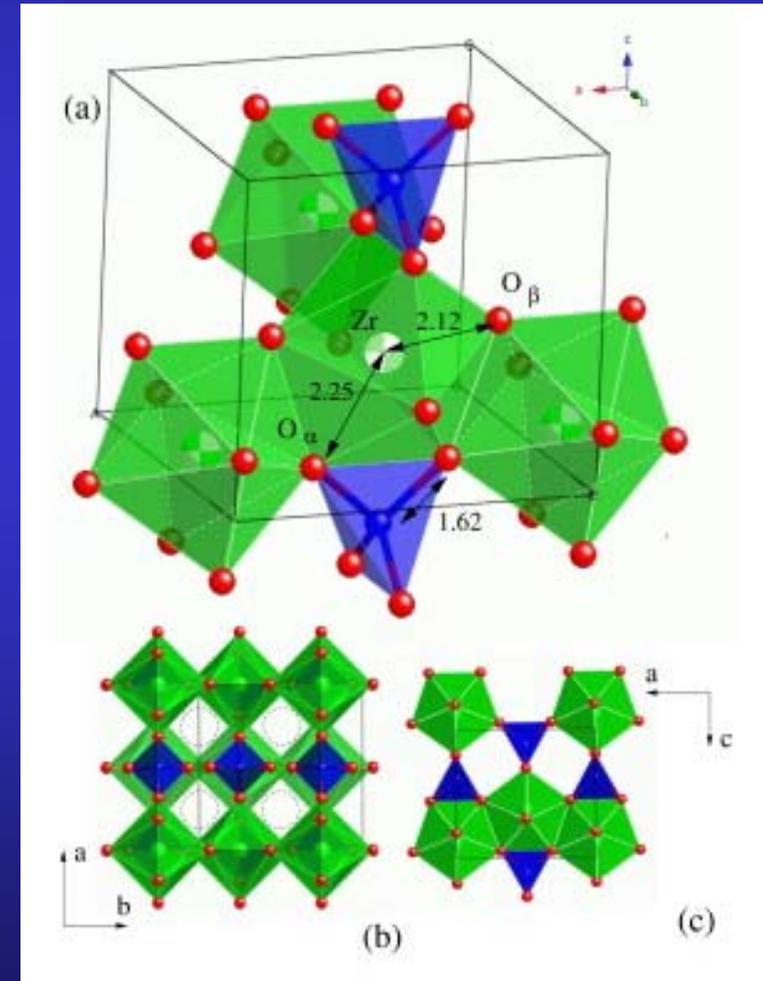
Intrinsic point defects and volume swelling in ZrSiO₄

- Radiation cascades & defect accumulation

$\sim 10^{21}$ defects/cm³

Si ...0.2% swelling!!

- Tetragonal $I4_1/amd$ space group
- BCC unit cell with four formula units.
- Structural parameters: a , c , u , v



Alternating SiO₄ tetrahedra & ZrO₈ dodecadeltahedra

Playing with the “concentration of defects”

Supercell approach:

$$n_x \times n_y \times n_z$$

Repetitions of the unit cell

(Periodic super-structure of defects)

1x1x1	(24 atoms)
1x1x2	(48 atoms)
2x2x1	(96 atoms)
2x2x2	(192 atoms)



Lattice relaxation

Low-concentration limit

atomic relaxation only!

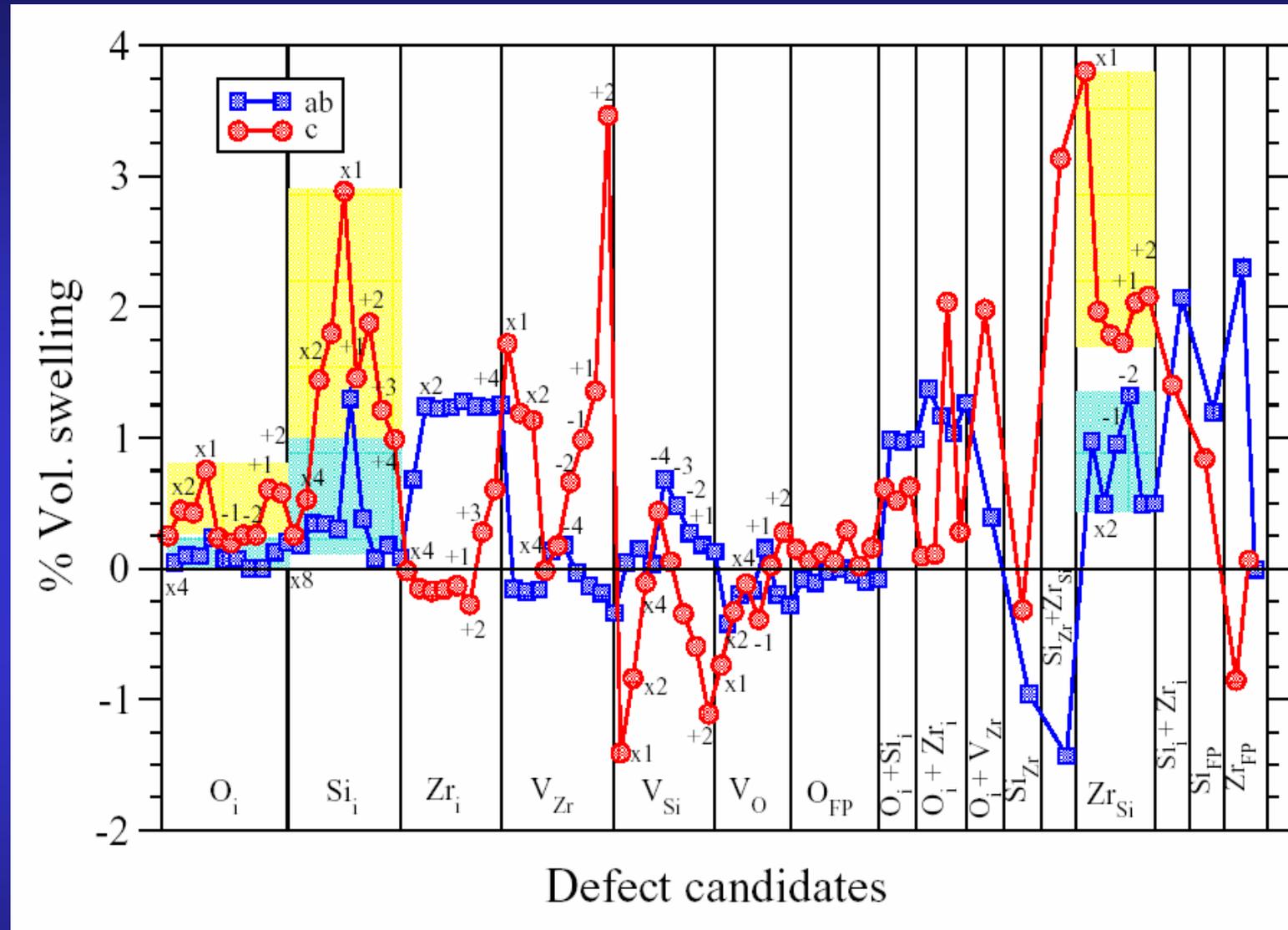


The catalogue of defects

- Interstitials: O_i , Si_i , Zr_i
- Vacancies: V_O , V_{Si} , V_{Zr}
- Antisite defects: Zr_{Si} & Si_{Zr}
- Frenkel pairs: O_{FP} , Si_{FP} & Zr_{FP}

Different charge-states:

-4, -3, -2, -1 | 0 | +1, +2, +3, +4



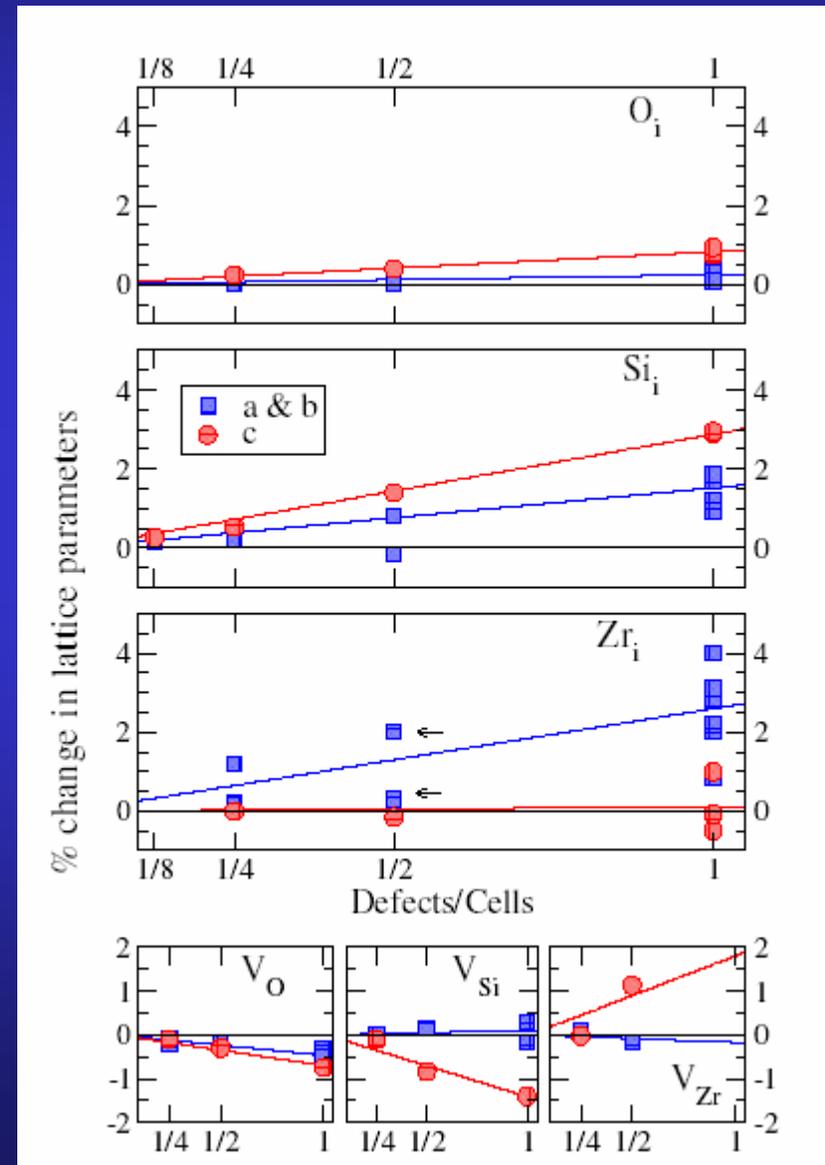
J. M. Pruneda, T. Archer, and E. Artacho *to be published in PRB*

Possible defects:

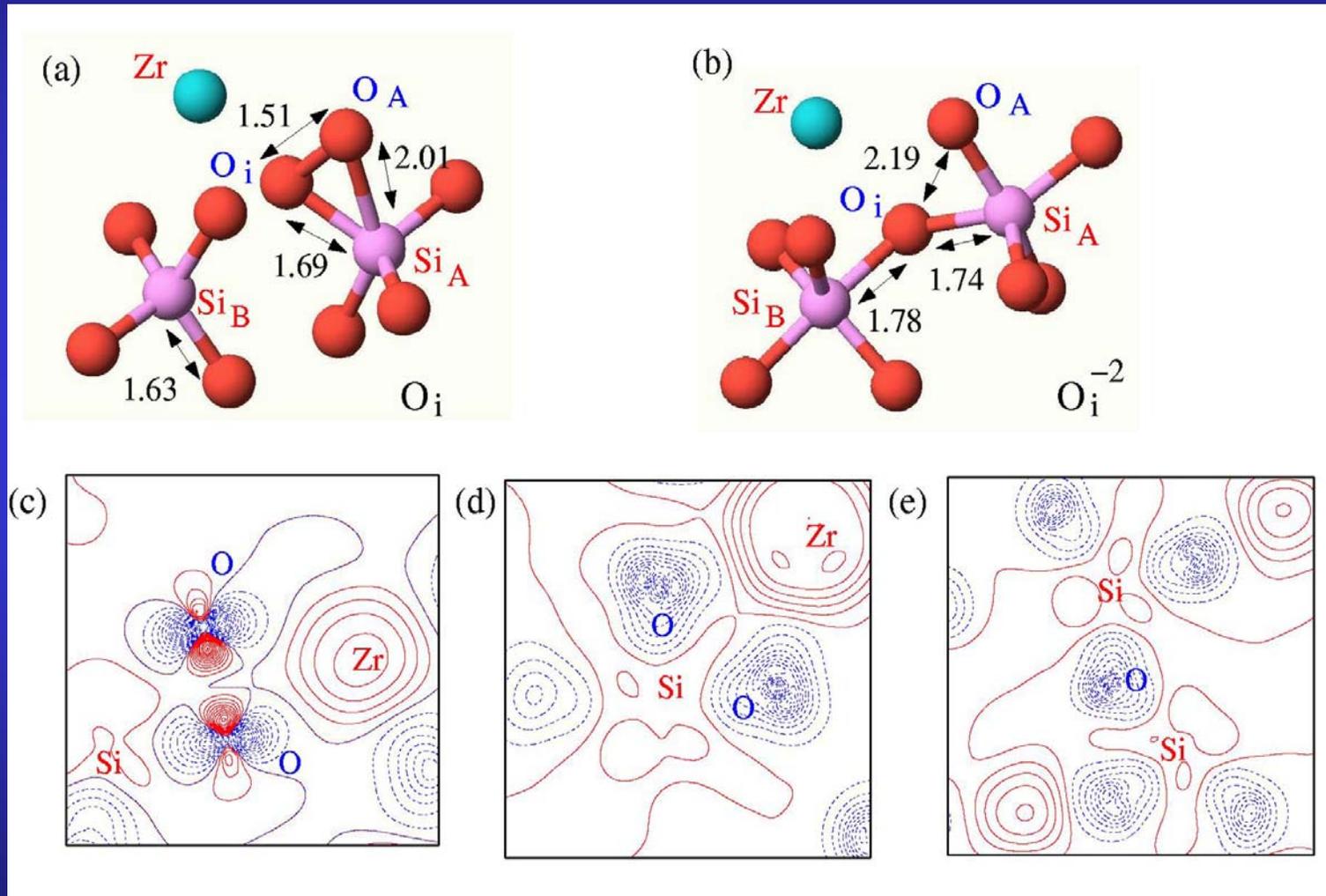
O & Si interstitials
Zr_{Si} anti-site

Swelling as a function of the concentration

- Almost linear behavior
- Considerable anisotropy
- Effect of disorder

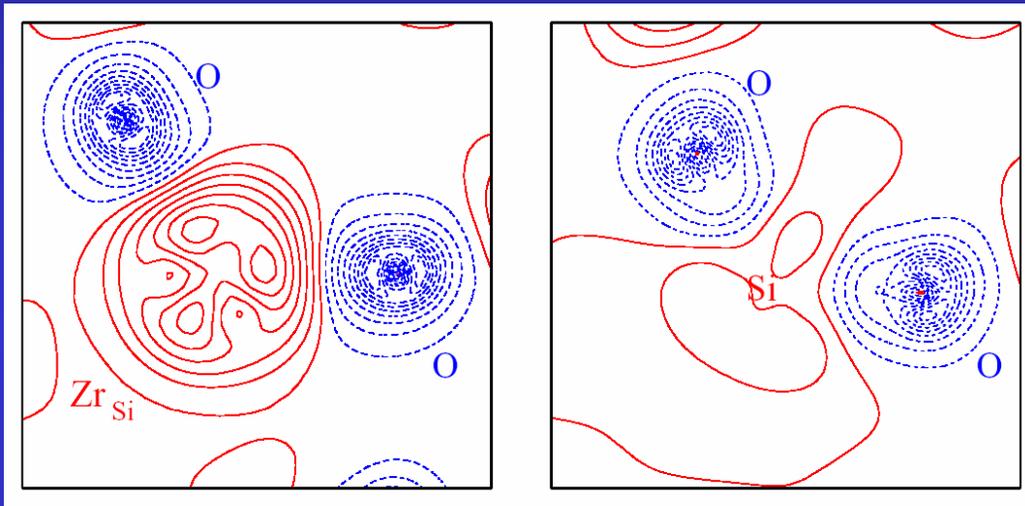


Oxygen interstitial



Neutral defect has a dumbbell structure similar to ZrO₂.

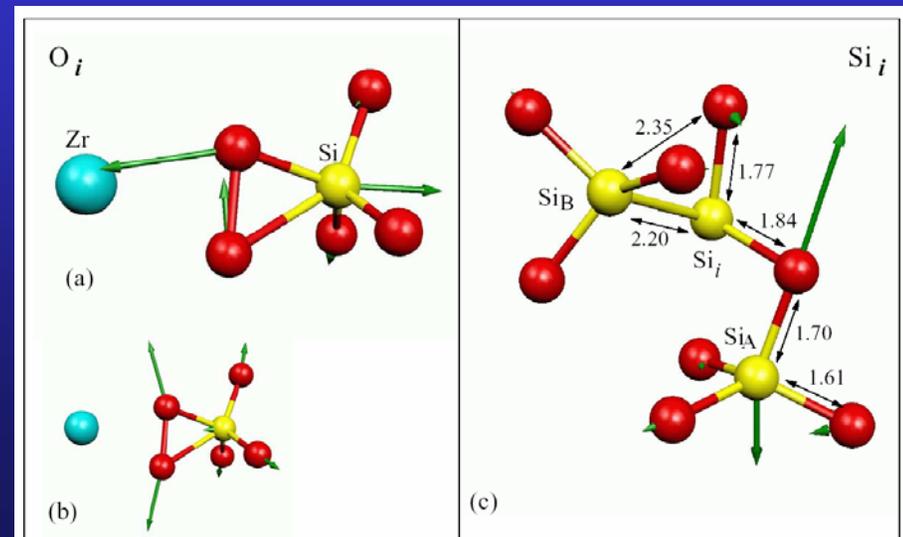
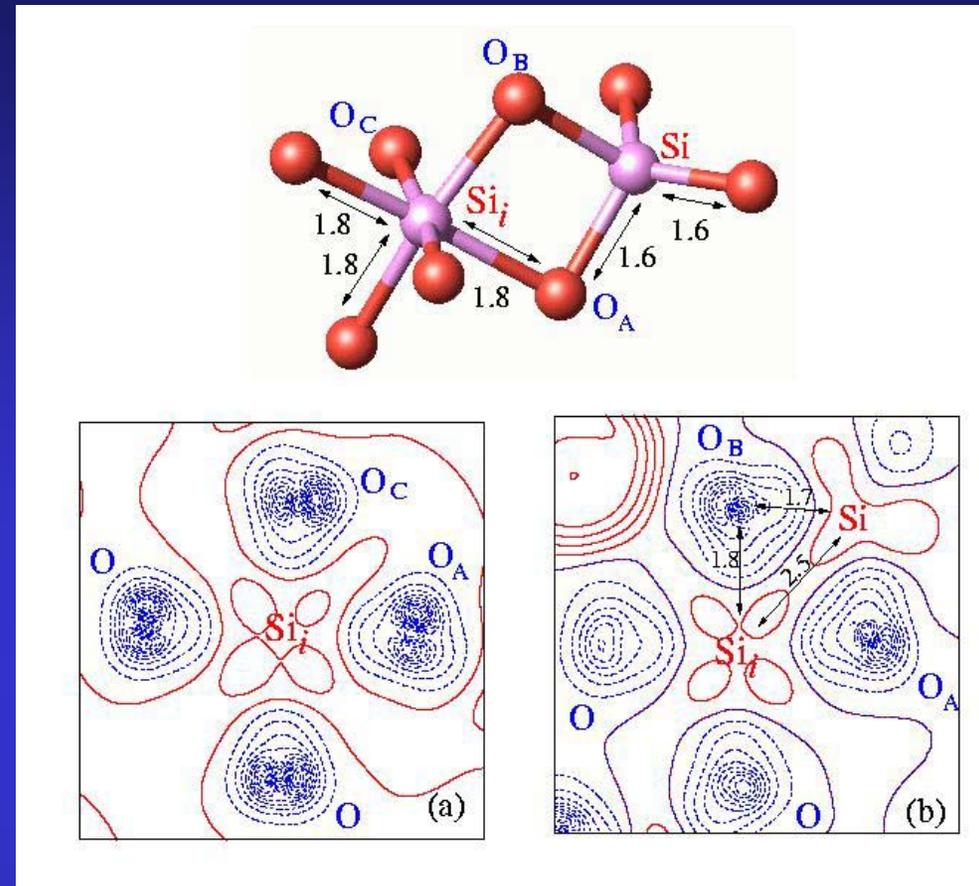
Silicon interstitial



Zr_{Si} anti-site

Local Vibrational Modes

IR active at 729 cm^{-1} (O_i)
 743 cm^{-1} (Si_i)

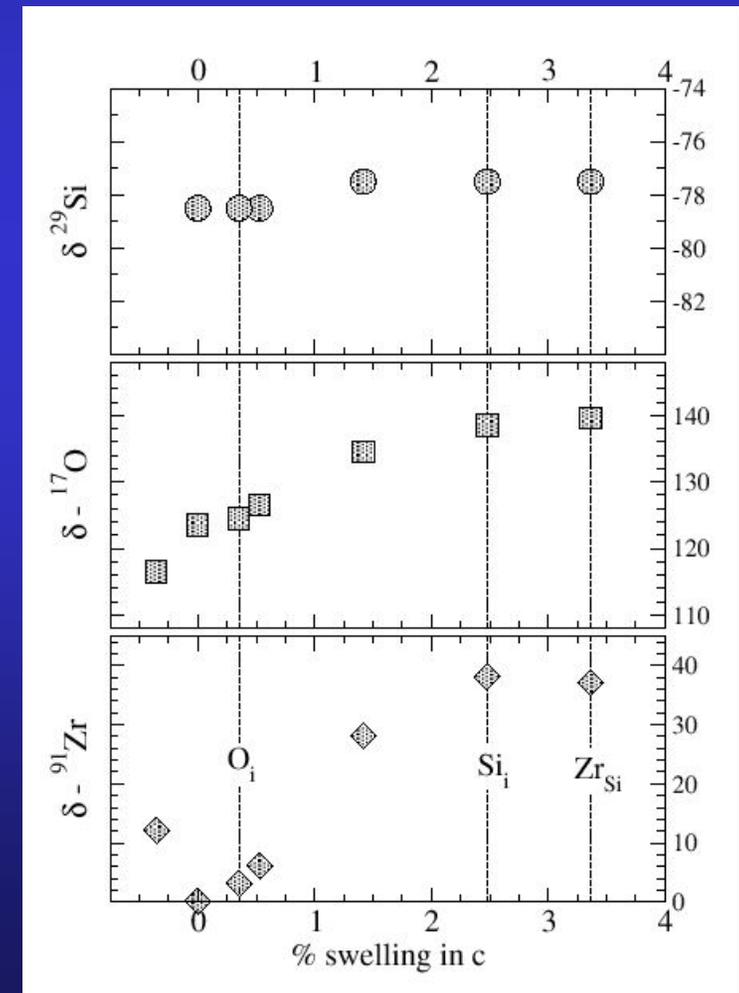
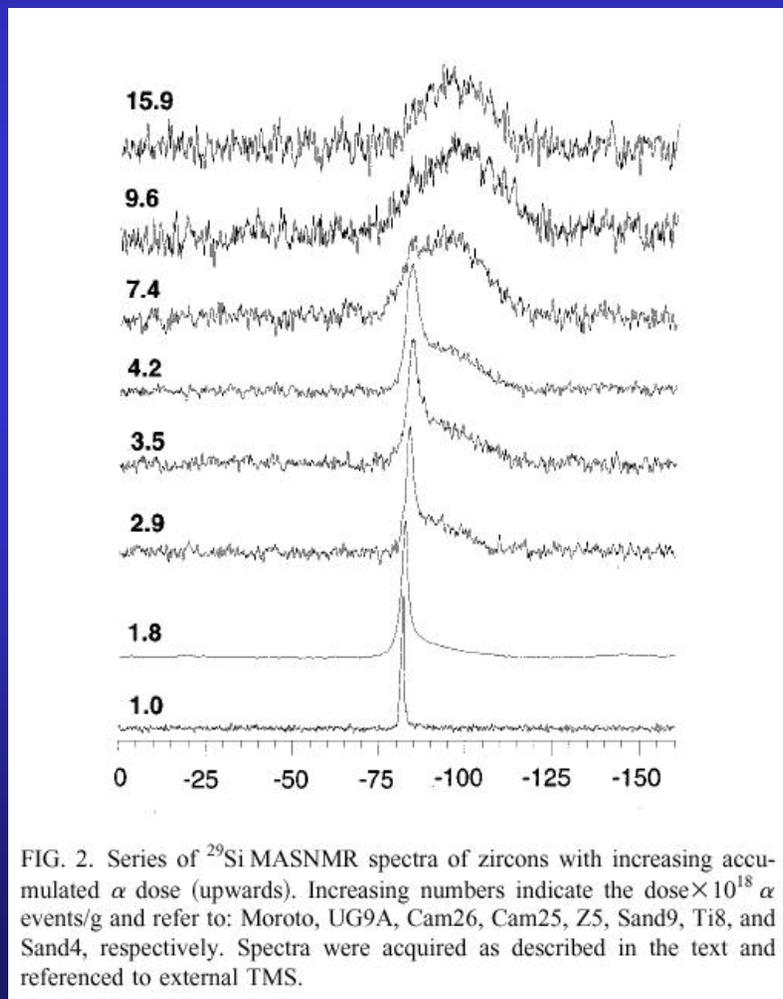


Simulation of NMR spectra

Combining methods: SIESTA+PARATEC

Experiments in damaged samples

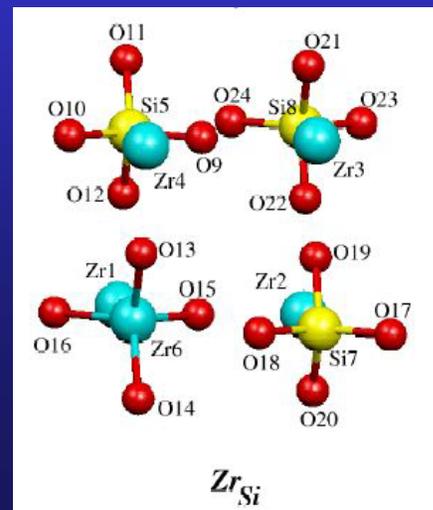
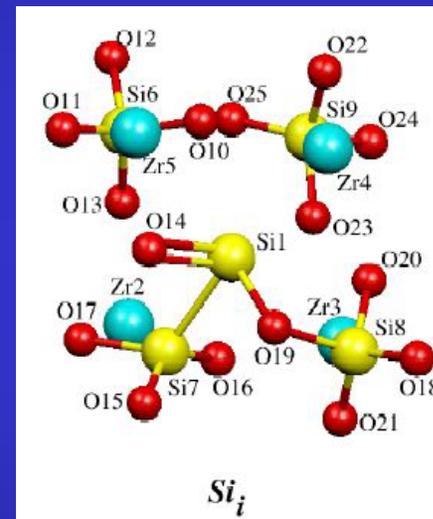
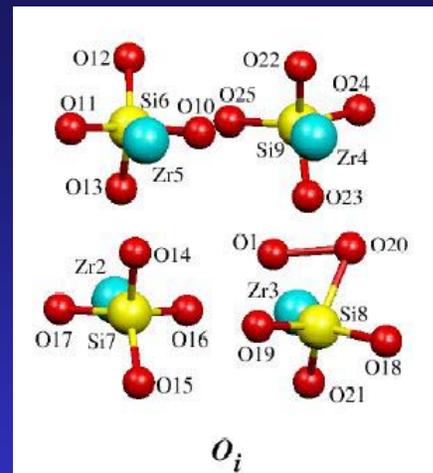
Chemical shifts due to swelling?



O: -170.1 (-170)
Si: -78.6 (-81.5)

	O_i	Si_i	Zr_{Si}
Si1		-7.5	
Si5			-82
Si6	-84	-76	
Si7	-81	-47	-82
Si8	-112	-75	-81
Si9	-86	-86	

	O_i	Si_i	Zr_{Si}
Zr1	-24		49
Zr2	-38	-54	-39
Zr3	65	-180	-65
Zr4	-30	-404	-39
Zr5		-121	
Zr6			-432



	O_i	Si_i	Zr_{Si}
O1	282		
O9			148
O10	196	117	148
O11	154	167	158
O12	199	224	158
O13	158	149	464
O14	164	244	464
O15	162	201	464
O16	169	322	464
O17	168	212	158
O18	183	178	158
O19	165	95	148
O20	178	203	148
O21	177	172	166
O22	176	161	166
O23	144	166	166
O24	166	188	166
O25	169	114	

Formation Energies

$$E_f(\alpha, q) = E(\alpha, q) - \sum_i \mu_i \cdot n_i + q(\mu_e + E_V)$$

$$\mu_{Zr} + \mu_{Si} + 4\mu_O = \Delta G_f(ZrSiO_4)$$

$$\mu_{Zr} + 2\mu_O \leq \Delta G_f(ZrO_2)$$

$$\mu_{Si} + 2\mu_O \leq \Delta G_f(SiO_2)$$

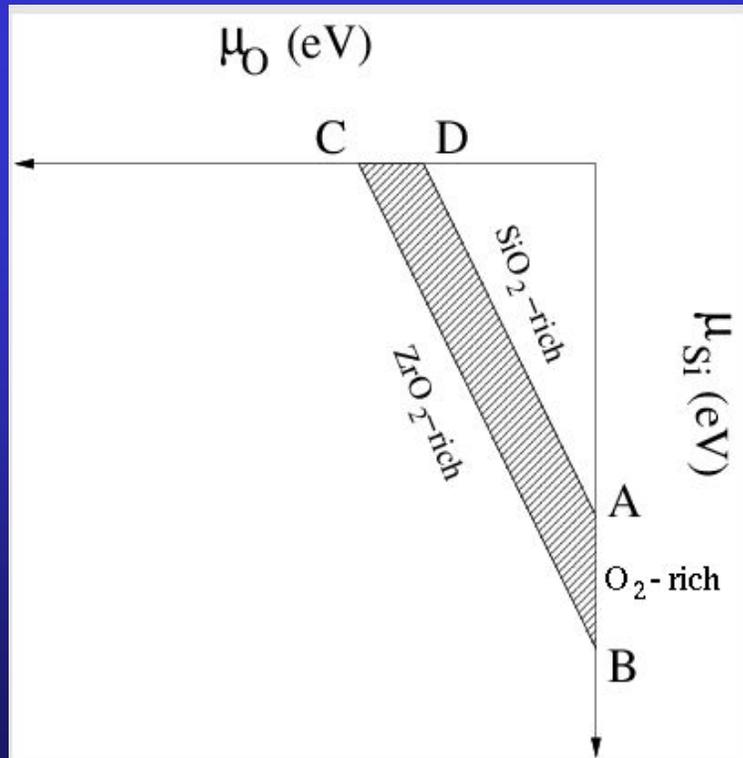
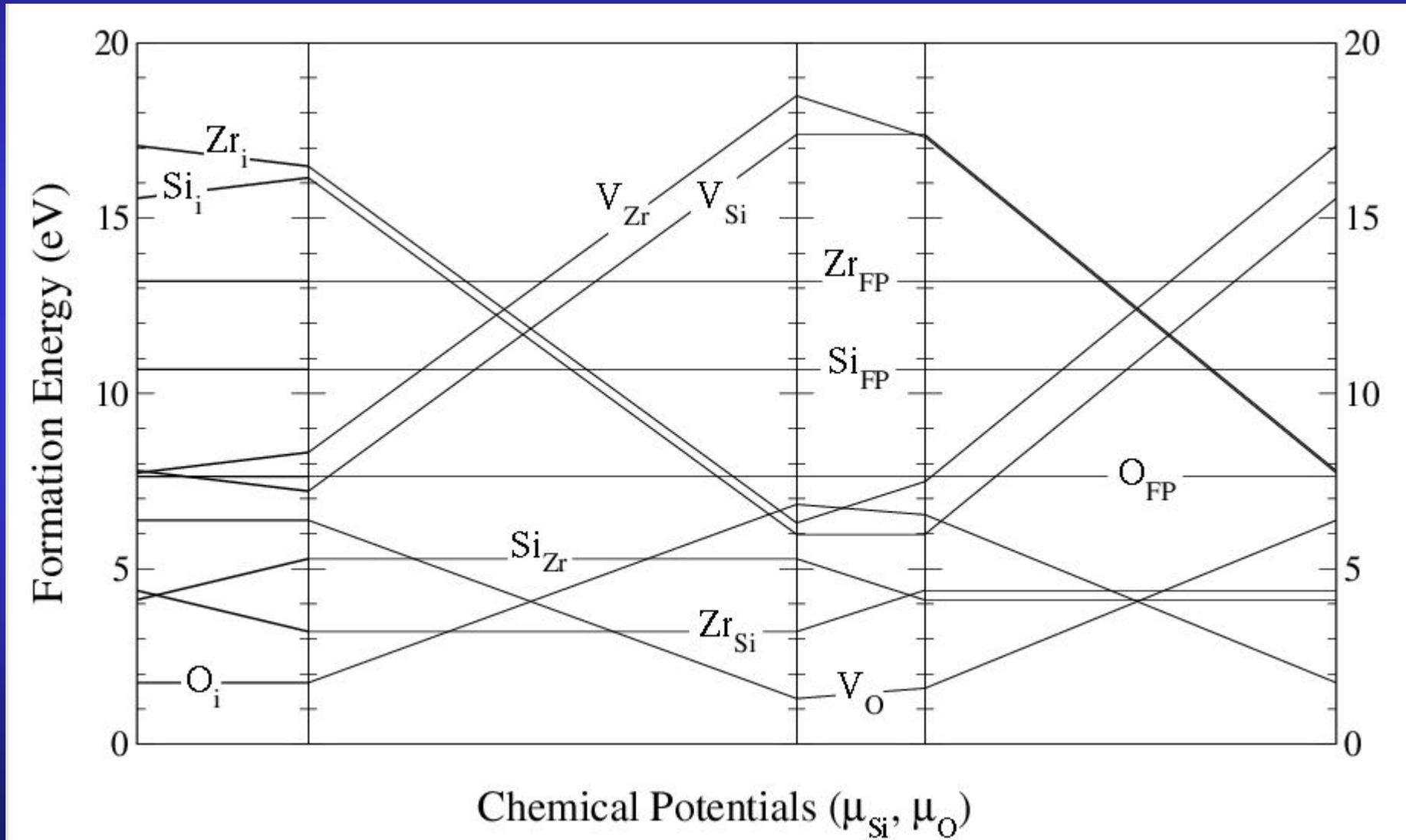


TABLE II: Calculated formation free energies (in eV) for $ZrSiO_4$, ZrO_2 , and SiO_2 .

Constituents	This work	Expt.
$Zr + O_2 \rightarrow ZrO_2$	-12.1	-11.5 ¹⁹
$Si + O_2 \rightarrow SiO_2$	-9.6	-9.8 ²⁰
$Zr + Si + 2 \times O_2 \rightarrow ZrSiO_4$	-22.3	-20.9 ²⁰

The stability triangle



Formation energies for neutral defects

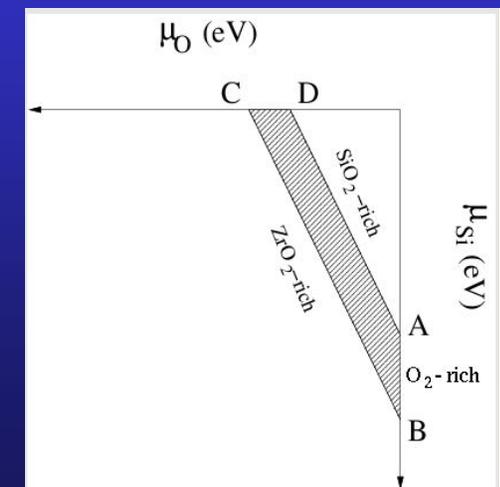
Defect α	24	48	96	192	Ref. [6]
O_i	1.7	1.7	2.0	1.1	1.7
Si_i	14.5	15.6	15.5	16.4	17.0
Zr_i		17.1	17.6	15.9	18.0
V_O	6.3	6.4	6.5	6.9	5.6
V_{Si}	3.6	7.8	8.5		5.8
V_{Zr}	7.4	7.7	7.9		5.9
O_{FP}		7.6			7.3
Si_{FP}		10.7			22.9
Zr_{FP}		13.2			24.0
Si_{Zr}		4.1			
Zr_{Si}	4.1	4.4		3.0	

Converged in size

*Values given
at the point "A"...*

More stable defects:

- Vacancies and interstitials of O
- Antisites.



Properties of charged defects

Negative-U behavior

- Gain in energy when a second electron/hole is captured.



- Strong ionic interaction for cation-defects (& charged FP)

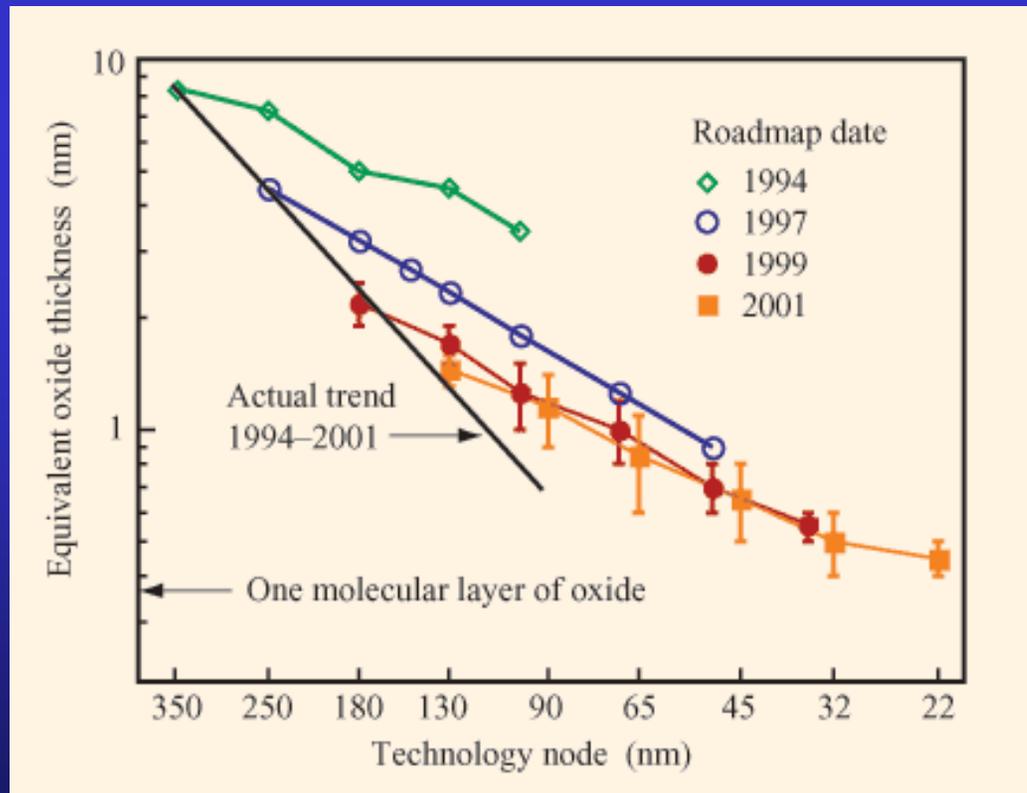


TABLE IV: Energies for defect reactions, obtained from the formation energies of isolated defects.

Reaction	Energy (eV)	
	high-C	low-C
$O_i^0 + O_i^{2-} \rightarrow 2O_i^-$	-0.4	-1.1
$V_O^0 + V_O^{2+} \rightarrow 2V_O^+$	-1.7	-1.0
$O_i^0 + V_O^0 \rightarrow O_i^- + V_O^+$	-0.5	-0.7
$O_i^- + V_O^+ \rightarrow O_i^{2-} + V_O^{2+}$	1.6	1.4
$O_i^0 + V_O^0 \rightarrow O_i^{2-} + V_O^{2+}$	1.1	0.7
$Si_i^0 + Si_i^{+2} \rightarrow 2Si_i^+$	-1.5	-1.0
$Si_i^0 + Si_i^{+4} \rightarrow 2Si_i^{2+}$	0.6	0.8
$Si_i^{+2} + Si_i^{+4} \rightarrow 2Si_i^{3+}$	-2.8	-2.6
$V_{Si}^0 + V_{Si}^{2-} \rightarrow 2V_{Si}^-$	0.3	
$V_{Si}^0 + V_{Si}^{4-} \rightarrow 2V_{Si}^{2-}$	1.7	
$Si_i^0 + V_{Si}^0 \rightarrow Si_i^- + V_{Si}^+$	3.0	
$Si_i^0 + V_{Si}^0 \rightarrow Si_i^{2-} + V_{Si}^{2+}$	7.2	
$Si_i^0 + V_{Si}^0 \rightarrow Si_i^{3-} + V_{Si}^{3+}$	8.3	
$Si_i^0 + V_{Si}^0 \rightarrow Si_i^{3-} + V_{Si}^{3+}$	12.0	
$Zr_i^0 + Zr_i^{+2} \rightarrow 2Zr_i^+$	-0.6	0.02
$Zr_i^0 + Zr_i^{+4} \rightarrow 2Zr_i^{2+}$	2.4	2.3
$Zr_i^{+2} + Zr_i^{+4} \rightarrow 2Zr_i^{3+}$	-0.2	0.1
$V_{Zr}^0 + V_{Zr}^{2-} \rightarrow 2V_{Zr}^-$	0.2	
$V_{Zr}^0 + V_{Zr}^{4-} \rightarrow 2V_{Zr}^{2-}$	1.3	
$Zr_i^0 + V_{Zr}^0 \rightarrow Zr_i^- + V_{Zr}^+$	4.3	
$Zr_i^0 + V_{Zr}^0 \rightarrow Zr_i^{2-} + V_{Zr}^{2+}$	9.0	
$Zr_i^0 + V_{Zr}^0 \rightarrow Zr_i^{3-} + V_{Zr}^{3+}$	11.7	
$Zr_i^0 + V_{Zr}^0 \rightarrow Zr_i^{4-} + V_{Zr}^{4+}$	14.3	

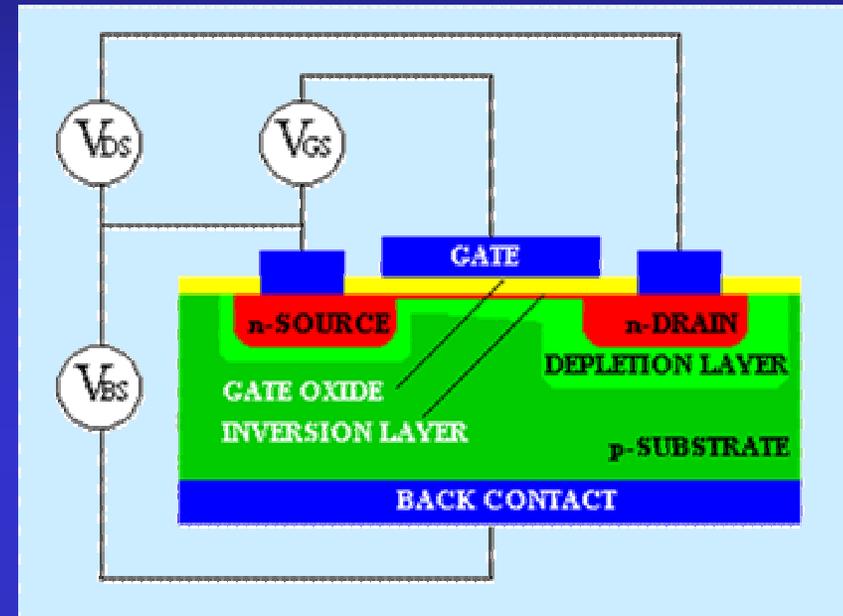
Alternative gate dielectrics

Device current drive is proportional to the oxide capacitance per unit area; thus, the best way to increase the drive current and thereby achieve high performance is to reduce the equivalent oxide thickness (EOT)



Scaling of the gate-oxide thickness:
Roadmaps vs. actual trends.

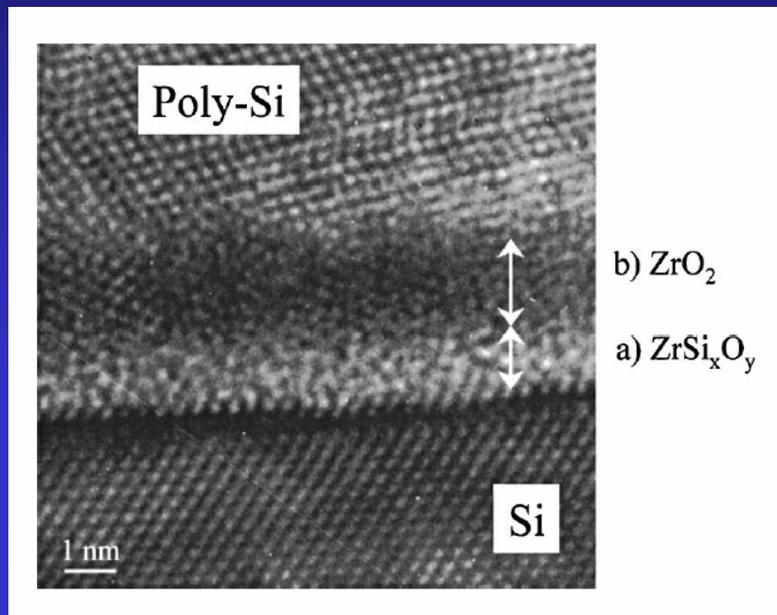
MOSFET



To reduce the gate leakage current:

1. the dielectric must be physically thick
2. it must have reasonably large band offsets to Si

Dielectric Constants of Zr Silicates

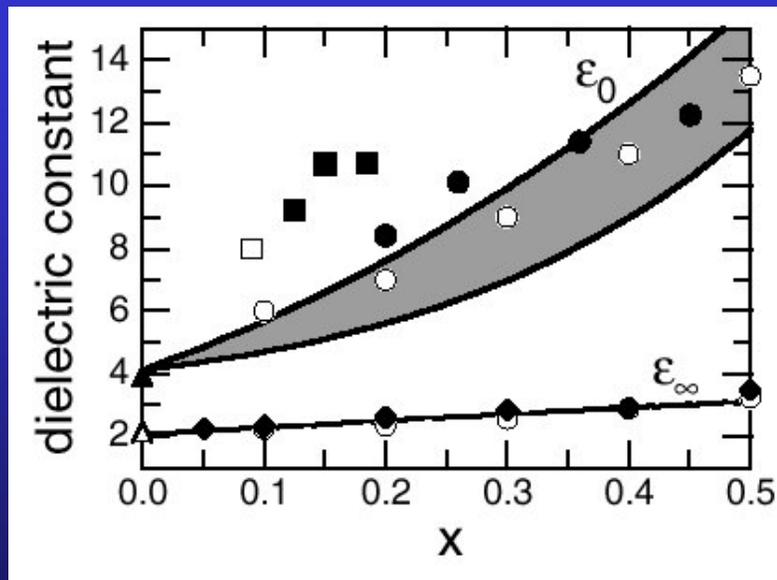


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and Development

Vol. 46, 2/3, 2002

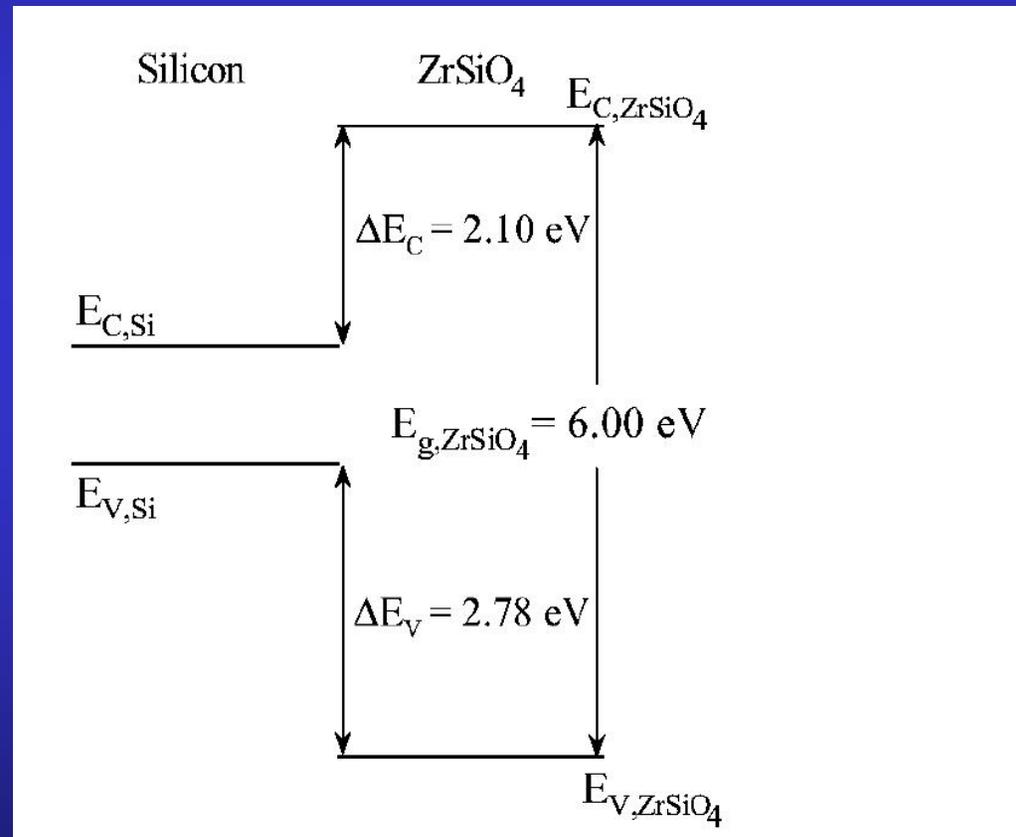
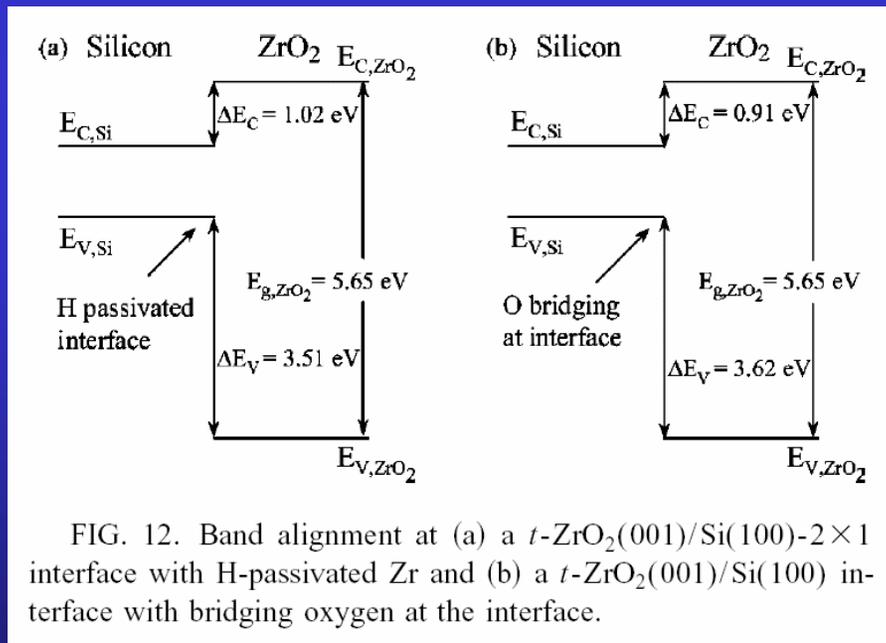
Vertically scaled MOSFET gate stacks and junctions: How far are we likely to go?

“Materials such as HfO_2 and ZrO_2 , having dielectric constants of about 15 and offsets of about 1.5 eV, have the potential of meeting the long-term leakage requirements. Their silicates have almost the same barrier height, nearly as high a dielectric constant, potentially lower charge levels, and much better thermal stability; thus, *they may be even better candidates.*”

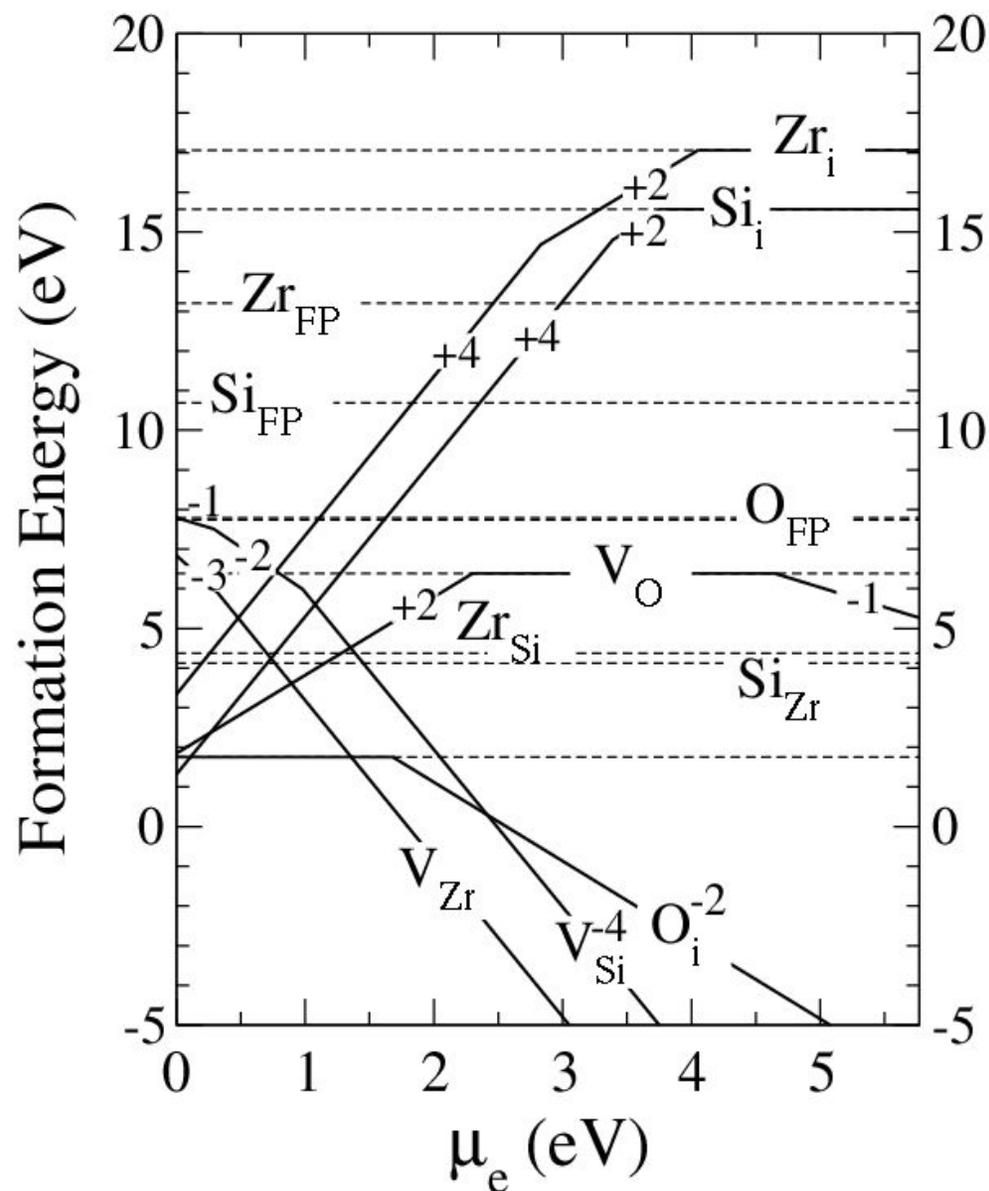


Band-Offsets

In the interface, it is required a high tunneling barrier to both electrons and holes.

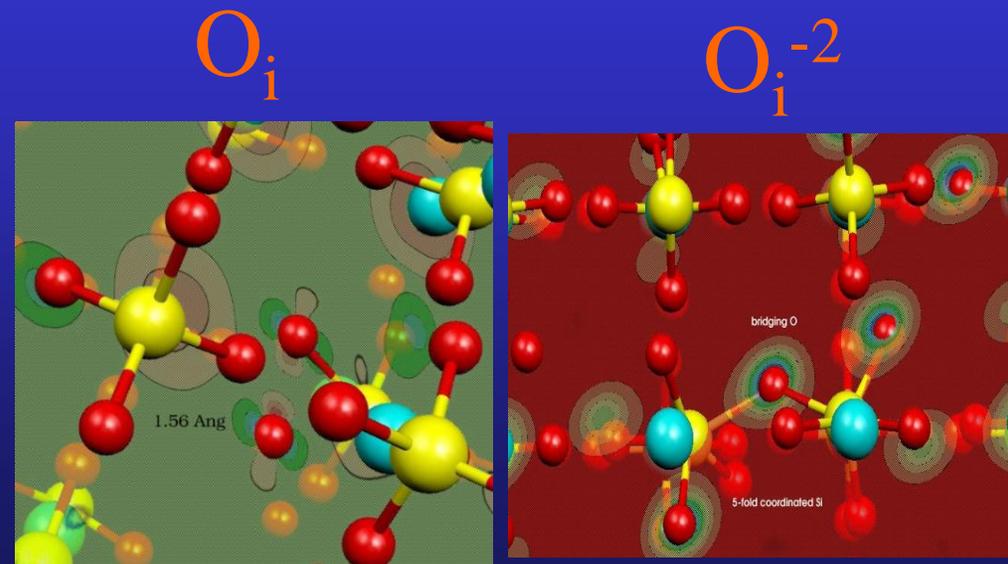


Charged defects!!



The dielectric cannot contain traps that would promote trap-assisted tunneling or locally uncompensated charges that would degrade channel mobility.

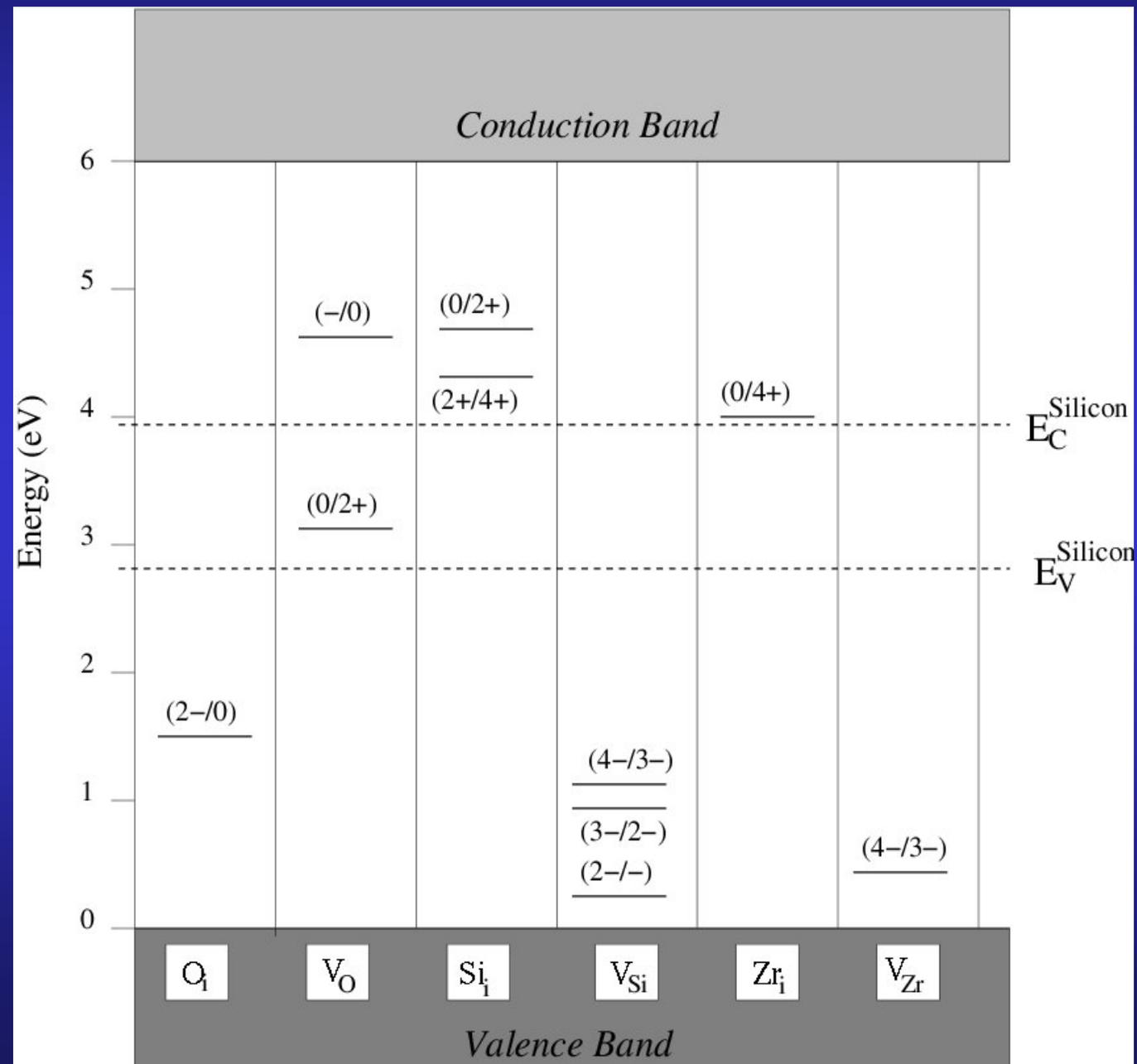
The conduction in the dielectric must be purely electronic, not ionic, and preferably by electrons only.



M

Transition Energies $E_{\alpha}(q/q') = [E_f(\alpha, q) - E_f(\alpha, q')]/(q' - q)$

- Acceptor levels for O_i , V_{Si} and V_{Zr} that can trap electrons from the bottom of the silicon valence band.
- Donor level of V_O can trap holes injected from silicon.



Conclusions

- *We have studied the effect of high concentration of point defects in ZrSiO₄.*
- *A roughly **linear dependence** between swelling and the defect concentration of periodically repeated defects was obtained.*
- *Based on experimental evidence of anisotropic lattice expansion, we have selected a **set of defects as good candidates to be responsible of the swelling: O_i, Si_i, and Zr_{Si}.***
- *Interstitials and vacancies of oxygen, and the antisite Zr_{Si} are the **most stable defects**.*
- *There is a strong tendency towards ionization, and a negative-U behavior is observed.*

Thanks to

Emilio Artacho, T. D. Archer,
Kostya Trachenko, Martin Dove,
Ian Farnan, Laurent Le Polles, Sharon Ashbrook, Jonathan Yates,
Greg Lumpkin, Susana Rios, Ming Zhang

