### Linking the Continuous and the Discrete

### Coupling Molecular Dynamics to Continuum Fluid Mechanics

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# **Outline**

### Introduction

- Motivation for coupling
- Molecular dynamics (MD)
- Computational fluid dynamics (CFD)

### . Mathematical Formulation for Coupling

- . Irving and Kirkwood (1950)
- . The control volume function
- Coupling using Hamilton's principle

### . Computational Developments

- . The molecular dynamics solver
- . The computational fluid dynamics solver
- The CPL\_Library (open source)





## Introduction



#### • Modern engineering problems require sub-continuum models

- Quantum mechanics is limited to very small systems
- Even Molecular dynamics is still prohibitively expensive
- Multi-scale coupling overcomes these limitations by linking to cheaper methods
- Quantum mechanics ↔ Molecular dynamics (MD) (Karplus, Levitt and Warshel)
- Multi-scale coupling has been employed since the 1970's (Curtin & Miller 2003) in solid mechanics modelling (e.g. for crack tips)
  - Essential to fully capture both the complicated detail in the crack and the impact on the wider system
  - Continuum ↔ Molecular mechanics (MM) ↔ Quantum mechanics (QM)
- . Classical coupling for fluids is less mature
  - Computational fluid dynamics (CFD) ↔ Molecular dynamics (MD) (O'Connell & Thompson 1995, Flekkøy et al 2000, Nie, Chen, E & Robbins 2004, Delgado-Buscalioni & Coveney, 2004)
  - Important for *e.g.* flow over carbon allotropes, biological systems, electronics, chemical reactions and combustion

- . Long term goal is to seamlessly link various scales of modelling
  - . Fine/coarse graining as required based on the problem of interest
  - Dynamic resource allocation and load balancing on multi-core architectures



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#### . Solving similar and relevant problems to quantum mechanics

- Relationship between the mathematical framework used to describe a continuous field and a discrete system (The Dirac delta function)
- Ensuring the dynamics agree in both systems in a physically meaningful manner
- Interfacing of computational solvers and data exchange
- Extend the range of quantum modelling, via MD-CFD coupling

# **Discrete Models (Molecular Dynamics)**

- . Discrete molecules in continuous space
  - Molecular position evolves continuously in time
  - Position and velocity from acceleration

$$egin{aligned} \dot{m{r}}_i &
ightarrow \dot{m{r}}_i \ \dot{m{r}}_i &
ightarrow m{r}_i(t) \end{aligned}$$



- . Acceleration obtained from forces
  - Governed by Newton's law for an N-body system
  - Point particles with pairwise interactions only

$$m_i \ddot{\boldsymbol{r}}_i = \mathbf{F}_i = \sum_{i \neq j}^N \boldsymbol{f}_{ij} \qquad \Phi(r_{ij}) = 4\epsilon \left[ \left(\frac{\ell}{r_{ij}}\right)^{12} - \left(\frac{\ell}{r_{ij}}\right)^6 \right]$$

## **Discrete Models (Molecular Dynamics)**





# **Continuum Field Equations**

- . Assumed continuous at every point in space
  - . Mass Conservation

$$\frac{\partial \rho}{\partial t} = -\boldsymbol{\nabla} \cdot \rho \boldsymbol{u}$$

. Momentum Balance (Newton's Law)

$$rac{\partial}{\partial t}
ho oldsymbol{u} + oldsymbol{
abla} \cdot 
ho oldsymbol{u} oldsymbol{u} = oldsymbol{
abla} \cdot oldsymbol{\Pi}$$

. Energy Conservation

$$rac{\partial}{\partial t}
ho \mathcal{E} = -oldsymbol{
abla}\cdot [
ho \mathcal{E}oldsymbol{u} + oldsymbol{\Pi}\cdotoldsymbol{u} + oldsymbol{q}]$$



Direct Numerical Simulation of Turbulent Couette Flow

# Mathematical Formulation for Coupling



## **Coupling Schematic**



## Irving and Kirkwood (1950)

 $\alpha(\boldsymbol{r}^{N},\boldsymbol{p}^{N}) f(\boldsymbol{r}^{N},\boldsymbol{p}^{N},t) d\boldsymbol{r}^{N} d\boldsymbol{p}^{N}$  $\alpha$ ; 15/0)=00 S(z-a) dx = 1  $\mathcal{S}(\mathbf{x},\mathbf{v})f(\mathbf{x},\mathbf{v})d_{\mathcal{H}}=f(\mathbf{v},\mathbf{v})_{\mathbf{v}}^{2}$ not been observed. If the preferred interpr NSPORT PROCESSE ect ouscevel. if the preferren interpretations the group of Raman hands, 769, 718, 819, the soluted to each other in much the same Steras fixs da + fias The probability distril TRANSPORT PROCESSES In Private sity of representative point to sature or nation are sature the spectra of CO3 and CS3. tion function (relative den-ars phase space) we denote as the expectation value of the dynamical valuable  $\mathcal{A} \Gamma P A$ 171.9 cm<sup>-1</sup>. The band at 809.7 cm<sup>-1</sup> has been interpre 17: 9 cm<sup>-4</sup>. The band at 300.7 cm<sup>-1</sup> has been interpreted as 22×394, the rather large difference between calculated and decoursed forements being standard to the communic unner as similar groups and 829 cm as 2X.995, the rather large difference between calculated and observed incomercy being accritic in the proximity is due A. Instances at House the based could be an  $f(\mathbf{R}_1, \cdots, \mathbf{R}_N; \mathbf{p}_1, \cdots,$ ACKROWLEDGI The writers wish to express their incented satisfying the normalization condit and observed incruency being ascrines, in the proximity to the A, fundamental. However, this hand could be an . ne writers wan to express their inconcentuation of C. F. Hammer and E. L du Pont de Nemaus an to the A. Iondamental. However, this band could be an upper-stage band corresponding to S19 cm<sup>-1</sup>. The interpretation of the faint sharp hand at 769.4 cm<sup>-1</sup> as an upper-stage band, 778+v<sup>-1</sup>-v<sub>-1</sub> scame more plauible than its interpretation as the B<sub>1</sub> combination, 2104-558 cm<sup>-1</sup>. The faint sharp Racean hand at \$25 8(1) 2= Sol TRACTOROL' PROCE Jerome Karle lor electron diffraction dats; and to Dr. E. K. Phyler and the National Bureau of Standards for  $\int id\mathbf{R}_1\cdots d\mathbf{R}_N dp_1\cdots dp_N = 1$ We shall use (2.7) to derive the equations of hydro-Averannize plausible than its interpretation as the B w combination, 2004-558 cm<sup>-1</sup>. The faint sharp Raman band at \$28.5 ment here been betweened on an unmarging band where  $d\mathbf{R}_t$  stands for a volume element in the conlata in the long wave-kngth region. 210+558 cm<sup>-1</sup>. The faint sharp Raman band at 835. cm<sup>-1</sup> has been interpreted as an upper-stage band. 2X a07 + v<sub>0</sub> − v<sub>0</sub>, rabter than as the R<sub>2</sub>, difference band 1540-510 cm<sup>-1</sup>, because the corresponding sum band water and stands for a volume element in the con-figuration space and dp a volume element in the two-non-time source of the ath weakers of schemes in time. (2.1) figuration space and dp, a volume element in the mo-mentany space of the 4th moderate f charges in time accounting on the well-become fine station III. STATISTICAL MICHANICAL EXPRESSIONS FOR DENSITIES menum space of the sun molecule. ; charges a according to the well-known Liouville equation The equations of hydrodynamics (1.1), (1.2), and (1.4) are concerned with densitive in ecdancy 3-space, e.g., mass density, momentum density, and energy density. We shall now express these as the expectation values of hydramical variables over an ensemble baving distribution function f. JUNE. 1950 VULUME 15. NUMBER 6 2:0  $\frac{df}{dt} = \sum_{k=0}^{N} \left[ -\frac{p_k}{m_k} \cdot \nabla \mathbf{R}_k f + \nabla \mathbf{R}_k U \cdot \nabla p_k f \right]$ The Statistical Mechanical Theory of Transport Processes. TV, The Equations of Hydrodynamics THE INCREAT OF CREMICAL PHYSICS where C is the potential energy of the entire gystem. Any (ymmetical variable,  $o(\mathbf{R}_1, \cdots, \mathbf{R}_r; p_1, \cdots, p_n)$ , her an expectation value given at time  $f(\mathbf{b}_1, \cdots, \mathbf{R}_r)$ . is the probability per unit volume that the kth molecules be at  $\mathcal{R}_{k}$  is  $\langle a_i f \rangle = \int_{a_{ij} f_{adj}} \int a \langle \mathbf{R}_{l_1} \cdots, \mathbf{p}_{l_j} \cdots \rangle f \langle \mathbf{R}_{l_1} \cdots, \mathbf{p}_{l_j} \cdots \rangle f \rangle$  $\int \cdots \int f(\mathbf{R}_1, \cdots, \mathbf{p}_k, \cdots, \mathbf{i}) d\mathbf{R}_1 \cdots$  $(\mathbf{R}_1, \dots, \mathbf{R}_l, \dots, \mathbf{r}_l)$  where the integration is over all position vectors ex-cept  $\mathbf{R}_i$  and over all momenta vectors. Introducing that the kth molecule is at the tribubility we unit velocues that the kth molecule is at the time time.  $\partial$  $\partial \alpha$ We thus denote by (a;j) the expectation value of a that the kth is induct of a and j inducts  $j \in \{1\}$  is indeed to be and j induct of a start  $j \in \{1\}$  is indeed to be and j induct of a start  $(a(R_k - \gamma);j)$ . Providing a does not descend on time embicitly, the  $\alpha$ ; 81  $\partial \eta$ this program is uncertain in a liquid, vantus attempts have been under to obrain a closed equation satisfies are immuly by the robulitity distillution function oper immuly by the robulity distillution function. Providing a close not depend on time explicitly, the rate of change of the expectation value of a k given by 81  $y^{(\alpha_i,f)} = \left\langle \alpha_i, \frac{\partial f}{\partial t} \right\rangle = \sum_{i=1}^{n} \left[ \left\langle \alpha_i, -\frac{p_{\lambda}}{m_{\lambda}}, \gamma_{m_{\lambda}, \ell} \right\rangle \right]$  $-\int \cdots \int \delta(R_{4}-t)/(R_{d_{1}}\cdots p_{d_{1}}\cdots p_{d_{1}}\cdots p_{d_{k}})dR_{t}\cdots dp_{d_{k}}$ generalization of the well-known Fokker-Planck equi HIS paper will be concerned with a der the equations of hydrodynamics from the prinkerentised antervos of assumptio Int (Re & Solt) generalization of the well-known Forker-Flanck equa-tion of stochastic theory, has been derived by Kj-L ations of hydrodynamics from the prin-classical statistical mechanics. In par-distantion of continuity, the equation of the base ation of energy transport will be above the state of the 100 DI SUCCIBEN: 40075; INS 1000 university of anti-wood by introducing the concepts of time smoothing and a fathing concepts of the second section has been These may I look Berlonger - mx prage The total that is density at r due to all asolecules is thus wood ny introducing use concepts as true smoothing and a triction constant. This latter equation has been ANO + (a; V& D. Voul) . (2.4) and a metuon constant. Los after equeton na ocen applied to obtain an expression for the creas tensor By tress feneral molecular variables. applied to obtain an expression for the stress unautified to obtain an expression for the stress unautified to be stress of third velocity and, thereby, esen applied in the space of R. 10-015=0 unon in grameris of must velocity and, incertify, ex-pressions (in terms of colorniar variables) (ar co rejents of above and volume viscosity.) The same Licular, motion, and the petic part (which derived. By so doing, the  $P(z; t) = \sum_{k=1}^{N} m_k (\delta(\mathbf{R}_k - z));$ density can be expressed in len oother term The stress tensor consists of a ot showr and volume viscosity. And same The m occurs in the kinetic theory of gases) a providing the integrat (dominant for a liquid) which will be expe 95 B actions of actionments emperatures VRACI () it ja .. quadrature involving the potential of interm the system is bounded or the use compensate gradient, rears to Eps. 1. (3.1) nd the locations of the quadrature involving the position of atternance to a terminate the density of pairs of molecules. The here the terminate the terminate the terminate termina as  $R_k \rightarrow \infty$ . Likewije, since  $\nabla R_k$ (2.5) the coefficient of thermal connote and the uenaty of pairs of molecules. The best current density is the sum of the ismiliar kinetic part by the matin vanishes; i.e., providi mentum pa, and since / falls of others are a current density is the sum of the immitar kinetic par-and a quarkature involving the potential of inter-medicatar force and the density and current density in the configuration ensure of a voir of motionale the enfor purposes of matheaiter lines. of sufficien ly ray of Green's theorem in the most  $\int p_{i\delta}(\mathbf{R}_{k}-z)/(\mathbf{R}_{i_{1}}\cdots;p_{i}\cdots;i)_{d}\mathbf{R}_{j}\cdots dp_{d'}$ an explicit expr opent, single phase malecular force and the density and current density in the configuration space of a pair of nulceulus. The re-sults were revolvably stated in the first article of this artist, when this defination are promised. To obtain explicit expressions for the pair probability density and probability current density one would in to an expression it We shall ass which interact simplicity, a single  $(\alpha; \nabla \mathbf{r}_{0}U, \nabla \mathbf{p}_{0}j) = -(\nabla \mathbf{r}_{0}U, \nabla \mathbf{r}_{0})$ Juide system consisting u noiection under central forces culy. It is not difficul ductivity."  $\int \cdots \int \delta(R_s - t) f(R_{l_s} \cdots; p_{l_s} \cdots; j) dR_1 \cdots dp_{l_s}$ Thus, (2.4) becomes toner central reversionly. It is not component or the treatment to a multiple component or <sup>1</sup> M. Dorth Ard, H. S. Green, Proc. Boy. Suc. A188, 19 ( Kithowood, Buch, and Turren, J. Chern. Phys. 17, 1998 (Starwood, Stankinst, and Levingine prevariation). J. editor evaluation sensitis a more precise comparations of the frait wheat description function. To obtain explicit expressions for the pair probability density and probability current density one would in subminde most to relate the thread to consider the  $\exists a; f = \tilde{\Sigma}$ (2.5) density and provability current density one would in principle need to solve the Linuville equation (Eq. (2.2)) for the probability distribution in Gibbs phase over and then perform encoded intermediate curve ·VRH Consequently,  $(p_{i}(R_{i} - r), f)$  is the product of this mean momentum by the probability per unit volume that the kth molecule to at r; i.e., it is the contribution of the kth molecule to the monentum per unit volume (mass current density). The local momentum density (2.3) for the probability distribution in Gross phase space and then perious repealed laterations. Since ·Vnaif) \* This work was supported by the U.S. ONE under Contract Manar 244 with the California Institute of The Souther. 1 G. Kirkwood, J. Chon. Phys. 14, 301 (1946) HEN or the era momentum per unit values (mass current density), "The local momentum density  $\sum \langle m_i \delta (\boldsymbol{r} - \boldsymbol{r}_i); f \rangle$  $\alpha = \sum m_i \delta \left( \boldsymbol{r} - \boldsymbol{r}_i \right)$  $\rho(\mathbf{r},t) = \mathbf{r}$ **Imperial College** 14 London

# Irving and Kirkwood (1950) cont.

. Density in the molecular system is defined

- -

$$\frac{\partial}{\partial t}\rho\left(\boldsymbol{r},t\right) = \frac{\partial}{\partial t}\sum_{i=1}^{N}\left\langle m_{i}\delta\left(\boldsymbol{r}-\boldsymbol{r}_{i}\right);f\right\rangle \qquad \qquad \frac{\partial\rho}{\partial t} = -\boldsymbol{\nabla}\cdot\rho\boldsymbol{u}$$

. Time evolution from the Irving and Kirkwood procedure

$$\frac{\partial}{\partial t} \left\langle \alpha; f \right\rangle = \sum_{i=1}^{N} \left\langle \frac{\boldsymbol{p}_{i}}{m_{i}} \cdot \frac{\partial \alpha}{\partial \boldsymbol{r}_{i}} - \boldsymbol{F}_{i} \cdot \frac{\partial \alpha}{\partial \boldsymbol{p}_{i}}; f \right\rangle \qquad \alpha = \sum_{i=1}^{N} m_{i} \delta \left(\boldsymbol{r} - \boldsymbol{r}_{i}\right)$$
$$\frac{\partial}{\partial t} \sum_{i=1}^{N} \left\langle m_{i} \delta \left(\boldsymbol{r} - \boldsymbol{r}_{i}\right); f \right\rangle = \sum_{i=1}^{N} \left\langle \frac{\boldsymbol{p}_{i}}{m_{i}} \cdot \frac{\partial}{\partial \boldsymbol{r}_{i}} m_{i} \delta \left(\boldsymbol{r} - \boldsymbol{r}_{i}\right) \right.$$
$$\left. - \boldsymbol{F}_{i} \cdot \frac{\partial}{\partial \boldsymbol{p}_{j}} m_{i} \delta \left(\boldsymbol{r} - \boldsymbol{r}_{i}\right); f \right\rangle$$
$$= -\frac{\partial}{\partial \boldsymbol{r}} \cdot \sum_{i=1}^{N} \left\langle \boldsymbol{p}_{i} \delta \left(\boldsymbol{r} - \boldsymbol{r}_{i}\right); f \right\rangle$$
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## **Mass Conservation Equation**

. Continuum mass conservation

$$rac{\partial 
ho}{\partial t} = - \boldsymbol{\nabla} \cdot 
ho \boldsymbol{u}$$

. Molecular Equivalent

$$\frac{\partial}{\partial t} \underbrace{\sum_{i=1}^{N} \left\langle m_i \delta\left(\boldsymbol{r} - \boldsymbol{r}_i\right); f \right\rangle}_{\rho} = -\boldsymbol{\nabla} \cdot \underbrace{\sum_{i=1}^{N} \left\langle \boldsymbol{p}_i \delta\left(\boldsymbol{r} - \boldsymbol{r}_i\right); f \right\rangle}_{\rho \boldsymbol{u}}$$

. Similarly for the time evolution of momentum (and beyond)

$$rac{\partial}{\partial t}
ho oldsymbol{u} = -oldsymbol{
abla}\cdot\left[
hooldsymbol{u}oldsymbol{u}+oldsymbol{\Pi}
ight]$$

# **Selecting Functions**

### . The Dirac delta selects molecules at a point

- Infinitely high, infinitely thin peak
- Equivalent to the continuum differential formulation at a point

$$\boldsymbol{\nabla} \cdot \boldsymbol{A} \equiv \lim_{V \to 0} \frac{1}{V} \oint_{S} \boldsymbol{A} \cdot d\mathbf{S} \qquad \rho(\boldsymbol{r}, t) = \sum_{i=1}^{N} \left\langle m_{i} \delta\left(\boldsymbol{r} - \boldsymbol{r}_{i}\right); f \right\rangle$$



- . In a molecular simulation r is never exactly equal to  $r_i$ 
  - Other difficulties with the Dirac delta function
- . Relaxed weighting functions
  - By Hardy(1981), Hoover (2009), Murdoch (2010) and others

$$\rho(\mathbf{r},t) \neq \sum_{i=1}^{N} \left\langle m_i \Delta \left( \mathbf{r} - \mathbf{r}_i \right); f \right\rangle$$

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x

 $x_i$ 

 $\Delta (x - x_i)$ 

# **The Control Volume (CV)**

#### . A finite volume with fluxes and forces acting over its surfaces

. Mass Conservation

$$\frac{\partial \rho}{\partial t} = -\boldsymbol{\nabla} \cdot \rho \boldsymbol{u}$$

$$\frac{\partial}{\partial t} \int_{V} \rho dV = -\oint_{S} \rho \boldsymbol{u} \cdot d\mathbf{S}$$

Momentum Balance (Newton's Law)

$$\frac{\partial}{\partial t}
ho \boldsymbol{u} + \boldsymbol{\nabla}\cdot 
ho \boldsymbol{u} \boldsymbol{u} = \boldsymbol{\nabla}\cdot \boldsymbol{\Pi}$$

$$\frac{\partial}{\partial t} \int_{V} \rho \boldsymbol{u} dV = -\oint_{S} \rho \boldsymbol{u} \boldsymbol{u} \cdot d\mathbf{S} + \mathbf{F}_{\text{surface}}$$

$$rac{\partial}{\partial t}
ho \mathcal{E} = -oldsymbol{
abla} \cdot [
ho \mathcal{E}oldsymbol{u} + oldsymbol{\Pi} \cdot oldsymbol{u} + oldsymbol{q}]$$

$$\begin{split} \frac{\partial}{\partial t} \int_{V} \rho \mathcal{E} dV &= -\oint_{S} \rho \mathcal{E} \boldsymbol{u} \cdot d\mathbf{S} \\ &-\oint_{S} \boldsymbol{\Pi} \cdot \boldsymbol{u} \cdot d\mathbf{S} + \mathbf{q} \cdot d\mathbf{S} \end{split}$$

# The Control Volume (CV)

- . Writing the molecular system in terms of control volumes
  - Mass

$$\rho = \sum_{i=1}^{N} m_i \delta\left(\boldsymbol{r} - \boldsymbol{r}_i\right) \qquad \qquad \int_{V} \rho dV = \sum_{i=1}^{N} m_i \int_{V} \delta\left(\boldsymbol{r} - \boldsymbol{r}_i\right) dV$$

. Momentum

$$\rho \boldsymbol{u} = \sum_{i=1}^{N} \boldsymbol{p}_{i} \delta\left(\boldsymbol{r} - \boldsymbol{r}_{i}\right)$$

$$\int_{V} \rho \boldsymbol{u} dV = \sum_{i=1}^{N} \boldsymbol{p}_{i} \int_{V} \delta \left( \boldsymbol{r} - \boldsymbol{r}_{i} \right) dV$$

• Energy

$$\rho \mathcal{E} = \sum_{i=1}^{N} e_i \delta\left(\boldsymbol{r} - \boldsymbol{r}_i\right) \qquad \qquad \int_{V} \rho \mathcal{E} dV = \sum_{i=1}^{N} e_i \int_{V} \delta\left(\boldsymbol{r} - \boldsymbol{r}_i\right) dV$$

# **Control Volume Function**

• The Control volume function is the integral of the Dirac delta function in 3 dimensions

$$\vartheta_{i} \equiv \int_{x^{-}}^{x^{+}} \int_{y^{-}}^{y^{+}} \int_{z^{-}}^{z^{+}} \delta(x_{i} - x) \delta(y_{i} - y) \delta(z_{i} - z) dx dy dz$$
  
=  $\left[ H(x^{+} - x_{i}) - H(x^{-} - x_{i}) \right]$   
×  $\left[ H(y^{+} - y_{i}) - H(y^{-} - y_{i}) \right]$   
×  $\left[ H(z^{+} - z_{i}) - H(z^{-} - z_{i}) \right]$ 

### • Or in words

if molecule is inside volume if molecule is outside volume

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 $\vartheta \equiv \begin{cases} 1\\ 0 \end{cases}$ 

For full details, please see E.R. Smith, D.M. Heyes, D. Dini, T.A. Zaki, Phys. Rev. E 85. 056705 (2012)



## **Derivatives yields the Surface Fluxes**

Taking the Derivative of the CV function

$$dS_{ix} \equiv -\frac{\partial \vartheta_i}{\partial x_i} = \left[\delta(x^+ - x_i) - \delta(x^- - x_i)\right] \\ \times \left[H(y^+ - y_i) - H(y^- - y_i)\right] \\ \times \left[H(z^+ - z_i) - H(z^- - z_i)\right]$$

Vector form defines six surfaces

$$d\mathbf{S}_i = i dS_{xi} + j dS_{yi} + k dS_{zi}$$

Or in words

 $d\mathbf{S}_i \equiv \begin{cases} \infty \\ 0 \end{cases}$ 

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if molecule on surface otherwise

For full details, please see

Phys. Rev. E 85. 056705 (2012)



# **Applying the Control Volume Function**

Molecular mass in a control volume can be defined •

$$\frac{\partial}{\partial t} \int_{V} \rho dV = \frac{\partial}{\partial t} \sum_{i=1}^{N} \left\langle \boldsymbol{m}_{i} \boldsymbol{\vartheta}_{i}; \boldsymbol{f} \right\rangle \qquad \qquad \frac{\partial}{\partial t} \int_{V} \rho dV = -\oint_{S} \rho \boldsymbol{u} \cdot d\mathbf{S}$$

Simple mathematical operations using the control volume • function  $\Lambda T$ λτ

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## **Reynolds' Transport Theorem**

#### . Mass, momentum and energy equations

. Mass Conservation

$$\frac{d}{dt}\sum_{i=1}^{N}m_{i}\vartheta_{i} = -\sum_{i=1}^{N}m_{i}\mathbf{v}_{i} \cdot d\boldsymbol{S}_{i}$$

Momentum Balance

$$\frac{d}{dt} \sum_{i=1}^{N} m_i \mathbf{v}_i \vartheta_i = -\sum_{i=1}^{N} m_i \mathbf{v}_i \mathbf{v}_i \cdot d\mathbf{S}_i + \frac{1}{2} \sum_{i,j}^{N} \boldsymbol{f}_{ij} \vartheta_{ij}$$

$$\begin{array}{l} \mathbf{Linergy \ Conservation} \\ \frac{d}{dt} \sum_{i=1}^{N} e_{i} \vartheta_{i} = -\sum_{i=1}^{N} e_{i} \mathbf{v}_{i} \cdot d\mathbf{S}_{i} \\ + \frac{1}{2} \sum_{i=1}^{N} \sum_{i \neq j}^{N} \frac{\mathbf{p}_{i}}{m_{i}} \cdot \mathbf{f}_{ij} \vartheta_{ij} \end{array}$$

$$\frac{\partial}{\partial t} \int_{V} \rho dV = -\oint_{S} \rho \boldsymbol{u} \cdot d\mathbf{S}$$

$$\frac{\partial}{\partial t} \int_{V} \rho \boldsymbol{u} dV = -\oint_{S} \rho \boldsymbol{u} \boldsymbol{u} \cdot d\mathbf{S}$$

$$+\mathbf{F}_{ ext{surface}}$$

$$\frac{\partial}{\partial t} \int_{V} \rho \mathcal{E} dV = -\oint_{S} \rho \mathcal{E} \boldsymbol{u} \cdot d\mathbf{S}$$

$$-\oint_{S} \mathbf{\Pi} \cdot \boldsymbol{u} \cdot d\mathbf{S} + \mathbf{q} \cdot d\mathbf{S}$$
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## **Reynolds' Transport Theorem**

- Mass, momentum and energy equations
  - Mass Conservation

$$\frac{d}{dt}\sum_{i=1}^{N} m_i \vartheta_i = -\sum_{i=1}^{N} m_i \mathbf{v}_i \cdot d\mathbf{S}_i$$

$$\frac{\partial}{\partial t} \int_{V} \rho dV = -\oint_{S} \rho \boldsymbol{u} \cdot d\mathbf{S}$$

Momentum Balance

$$\frac{d}{dt} \sum_{i=1}^{N} m_i \mathbf{v}_i \vartheta_i = -\sum_{i=1}^{N} m_i \mathbf{v}_i \mathbf{v}_i \cdot d\mathbf{S}_i + \frac{1}{2} \sum_{i,j}^{N} \boldsymbol{f}_{ij} \vartheta_{ij}$$

$$\frac{\partial}{\partial t} \int_{V} \rho \boldsymbol{u} dV = -\oint_{S} \rho \boldsymbol{u} \boldsymbol{u} \cdot d\mathbf{S}$$

 $+\mathbf{F}_{\mathrm{surface}}$ 



. Momentum Balance



## **The Pressure Tensor**

 Continuum Control Volume equations in terms of the pressure tensor

$$\frac{\partial}{\partial t} \int_{V} \rho \boldsymbol{u} dV = -\oint_{S} \rho \boldsymbol{u} \boldsymbol{u} \cdot d\mathbf{S} + \mathbf{F}_{\text{surface}} - \frac{\partial}{\partial r} \cdot \int_{V} \boldsymbol{\Pi} dV$$
$$-\oint_{S} \boldsymbol{\Pi} \cdot d\mathbf{S}$$

Molecular Control Volume equations in terms of the pressure
 tensor
 Volume Average

Volume Average (Lutsko 1988 & Cormier et al 2001)

$$\frac{d}{dt}\sum_{i=1}^{N}m_{i}\mathbf{v}_{i}\vartheta_{i} = -\sum_{i=1}^{N}m_{i}\mathbf{v}_{i}\mathbf{v}_{i} \cdot d\mathbf{S}_{i} + \underbrace{\frac{1}{2}\sum_{i,j}^{N}\boldsymbol{f}_{ij}\vartheta_{ij}}_{i,j} \qquad -\frac{\partial}{\partial\boldsymbol{r}} \cdot \underbrace{\frac{1}{2}\sum_{i,j}^{N}\boldsymbol{f}_{ij}\boldsymbol{r}_{ij}\int_{0}^{1}\vartheta_{s}ds}_{-\frac{1}{4}\sum_{i,j}^{N}\boldsymbol{\varsigma}_{ij} \cdot d\mathbf{S}_{ij}}$$

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Method of planes (Todd et al 1995, Han & Lee 2004)



- . We now have an equivalent description in both regions
  - Momentum or stresses inside an arbitrary control volume in both domains

$$\int_{V} \rho u dV = \sum_{i=1}^{N} m_{i} \dot{\boldsymbol{r}}_{i} \vartheta_{i}$$
$$\oint_{S} \boldsymbol{\Pi} \cdot d\boldsymbol{S} = \frac{1}{4} \sum_{i,j}^{N} \boldsymbol{\varsigma}_{ij} \cdot d\mathbf{S}_{ij}$$

**State Coupling :** O'Connell & Thompson (1995), Nie, Chen, E & Robbins (2004)









## **Constrained Control Volume**

### • Non-unique solution

- Continuum field properties must specify N molecules
- . Hamilton's principle (subject to a constraint)
  - . As close as possible to the true trajectory
  - Used in the first fluids coupling scheme (O'Connell and Thompson 1995)

$$\delta A_c = \delta \int_a^b \left( \mathcal{L} + \boldsymbol{\lambda} \cdot \boldsymbol{g} \right) dt = 0$$

- But now we want to apply a constraint <u>localised</u> using the control volume function
  - CV function takes care of the localisation for us

$$g(\boldsymbol{q}_i, \dot{\boldsymbol{q}}_i) = \sum_{i=1}^{N} m_i \dot{\boldsymbol{q}}_i \vartheta_i - \int_{V} \rho \boldsymbol{u} dV = 0$$

## **Principle of least action**

### . The Euler-Lagrange equation is applicable

• As the constraint is semi-holonomic (Flannery 2004,2011)

$$\frac{d}{dt}\frac{\partial \mathcal{L}_c}{\partial \dot{q}_i} - \frac{\partial \mathcal{L}_c}{\partial q_i} = 0 \qquad \qquad \mathcal{L}_c = \mathcal{L} + \boldsymbol{\lambda} \cdot \boldsymbol{g}$$

• Written in terms of canonical momentum and its time derivative

$$p_i = \frac{\partial \mathcal{L}_c}{\partial \dot{q}_i}$$
$$\dot{p}_i = \frac{\partial \mathcal{L}_c}{\partial q_i}$$

- . To give equations of motions
  - Applying the MD boundary condition based on Hamilton's principle
  - Localised to a region in space (courtesy of the CV function)
  - No energy added to the system when applied correctly (i.e. with the CV function)



# **Coupling Results**



# **Computing Developments**



## **Molecular Dynamics Solver**

- New parallel molecular dynamics (MD) code
  - Fortran 90 (including 'modern' 2003 features and python wrappers)
  - Optimised for non-equilibrium molecular dynamics and linking to CFD
- A range of verification tests using experimental data and literature benchmarks
  - . Phase diagrams
  - Radial Distribution Function (related to the scattering function)



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# **MD Computing – Serial optimisations**

- . All pairs simulation uses local cell/neighbour lists
  - N<sup>2</sup> calculation reduced to order N
  - . Linked lists are used in order to manipulate data
  - Result in non-contiguous data access
- . Hilbert curve sorting improves cache efficiency of operations
  - Improvement becomes greater on larger systems
  - Some tuning required

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- . Heaviside function implemented in assembly language
  - cmplesd xmm1,xmm0 #Check input less than 0.0; true=zeros, false=ones
    - movsd xmm0,xmm2 #Save 1.0 into xmm0
    - andpd xmm0,xmm1 #AND 1.0 with all zeros/ones



# **MD Computing – Serial optimisations**



# **MD Computing – Parallel optimisations**

- Localisations lends itself to parallel computing using MPI
  - Spatial decomposition employed
  - . Halo cells (ghost molecules) are used to link adjacent regions



- Halo exchange of variable amounts of data
  - MPI\_Send, MPI\_Probe and MPI\_Recv employed
- . CUDA implementation developed but found to be too inflexible to justify the speedup (especially in parallel)
  - 30x speedup reported if entire code on GPU (Anderson et al 2008)
  - Speedup negated by transfer, multi-GPU implementation challenging

# **MD Computing – Parallel optimisations**

- Strong scaling vs 1 core with 3,322,336 molecules
  - On HECToR and Imperial's supercomputers CX1/CX2
  - Efficiency of 90% when comparing 1024 cores to 8 cores



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# **Computational Fluid Dynamics**

### • Fortran finite volume (FV) Direct Numerical Simulation (DNS)

- Highly optimised algorithm used in simulation of turbulence
- Fully parallelised using MPI (halo cells) with good scalability in benchmark tests
- Extensively tested and verified (Zaki & Durbin, 2005, 2006)



Number of cores

## **CPL\_LIBRARY** Overview

### Based on the MPI module

- Designed in collaboration with Numerical Algorithms Group (NAG)
- Lightweight and efficient library to preserve the scaling of the two codes
- API using pure Fortran functions, unit tested and inclusive of error checking

### Framework is general

- Exchange any arbitrary data arrays per continuum cell
- Allows, in principle, the coupling of any continuum code to any molecular code
- MPMD implementation enforces separate scope of the two solvers





# **CPL\_LIBRARY Key Routines**

### Setup

- CPL\_create\_comm Split intra-comms and create inter-comms
- CPL\_CFD\_init / CPL\_MD\_init –Define mapping between domains. Every processor stores a copy of all mappings, communicators and both solvers processor topology

Exchange

- CPL\_send / CPL\_recv Send and receive data between overlapping processors on the coupling interface
- CPL\_gather / CPL\_scatter Gather/scatter operations implemented efficiently on dedicated communicators linking coupled processors

### • Enquiry

- CPL\_get Return copies of protected library internal data
- CPL\_cart\_coord Coordinate of any processor on either solver
- CPL\_COMM\_rank Rank of processor in specified communicator
- CPL\_extents extents of cells on current processor



## **CPL\_LIBRARY Schematic**



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## **CPL\_Library results**



## **CPL\_Library results**



# Conclusions

### • Multi-scale coupling is essential for modern engineering

• Implementation of molecular dynamics (and quantum mechanics) where essential and a continuum model to extend to engineering scales

### . Consistent framework for coupling fluid descriptions

- Control volume (CV) function expresses continuum and discrete systems in an equivalent formulation (rigours backing to existing coupling)
- Hamilton's principle, with the CV function, provides a constraint which ensures the descriptions in both domains agree
- Verified using test cases with known analytical solutions

### . Computational developments

- Extensive serial and parallel optimisations applied to the molecular dynamics solver
- CPL\_library is an open source library to facilitates the exchange of data between two coupled codes (<u>https://code.google.com/p/cpl-library/</u>)
- · Verified with test cases on a range of processor topologies

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## **Extra Material**



. Momentum Balance







#### **Momentum Balance**

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#### . Momentum Balance









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# **Control Volume Function (revisited)**

• The Control volume function is the integral of the Dirac delta function in 3 dimensions

$$\vartheta_i \equiv \int_V \delta(\mathbf{r} - \mathbf{r}_i) \, dV$$
  
=  $\begin{bmatrix} H(x^+ - x_i) - H(x^- - x_i) \end{bmatrix}$   
×  $\begin{bmatrix} H(y^+ - y_i) - H(y^- - y_i) \end{bmatrix}$   
×  $\begin{bmatrix} H(z^+ - z_i) - H(z^- - z_i) \end{bmatrix}$ 

• Replace molecular position with equation for a line

$$m{r}_i 
ightarrow m{r}_i - s m{r}_{ij}$$



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For full details, please see E.R. Smith, D.M. Heyes, D. Dini, T.A. Zaki, Phys. Rev. E 85. 056705 (2012)

# **Control Volume Function (revisited)**

• The Control volume function is the integral of the Dirac delta function in 3 dimensions

$$\vartheta_s \equiv \int_V \delta(\mathbf{r} - \mathbf{r}_i + s\mathbf{r}_{ij})dV = \begin{bmatrix} H(x^+ - x_i + sx_{ij}) - H(x^- - x_i + sx_{ij}) \end{bmatrix} \\ \times \begin{bmatrix} H(y^+ - y_i + sy_{ij}) - H(y^- - y_i + sy_{ij}) \end{bmatrix} \\ \times \begin{bmatrix} H(z^+ - z_i + sz_{ij}) - H(z^- - z_i + sz_{ij}) \end{bmatrix}$$

. Length of interaction inside the CV

$$\ell_{ij} = \int_{0}^{1} \vartheta_s ds$$



## **Derivatives Yield the Surface Forces**

. Taking the Derivative of the CV function

$$\frac{\partial \vartheta_s}{\partial x} \equiv \left[ \delta(x^+ - x_i + sx_{ij}) - \delta(x^- - x_i + sx_{ij}) \right] \\ \times \left[ H(y^+ - y_i + sy_{ij}) - H(y^- - y_i + sy_{ij}) \right] \\ \times \left[ H(z^+ - z_i + sz_{ij}) - H(z^- - z_i + sz_{ij}) \right]$$

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• Surface fluxes over the top and bottom surface

$$dS_{xij} \equiv \int_{0}^{1} \frac{\partial \vartheta_s}{\partial x} ds = dS_{xij}^{+} - dS_{xij}^{-}$$

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$$dS_{xij}^{+} = \frac{1}{2} \underbrace{\left[sgn(x^{+} - x_{i}) - sgn(x^{+} - x_{j})\right]}_{MOP} S_{xij}$$

# **More on the Pressure Tensor**

#### . Extensive literature on the form of the molecular stress tensor

- No unique solution Schofield, Henderson (1988)
- Two key forms in common use Volume Average (Lutsko, 1988) and Method of Planes (Todd et al 1995)
- . Link provided between these descriptions
  - Through formal manipulation of the functions
  - Exposes the relationship between the molecular stresses and the evolution of momentum
- In the limit the Dirac delta form of Irving and Kirkwood (1950) is obtained
  - This suggests the same limit is not possible in the molecular system
  - Arbitrary stress based on the volume of interest

![](_page_54_Figure_10.jpeg)

## **Moving reference frame**

- Why the continuum form of Reynolds' transport theorem has a partial derivative but the discrete is a full derivative
  - Eulerian mass conservation

$$\frac{d}{dt}\sum_{i=1}^{N} m_i \vartheta_i = -\sum_{i=1}^{N} m_i \mathbf{v}_i \cdot d\mathbf{S}_i$$

$$\frac{\partial}{\partial t} \int_{V} \rho dV = -\oint_{S} \rho \boldsymbol{u} \cdot d\mathbf{S}$$

Lagrangian mass conservation

$$\frac{d}{dt}\sum_{i=1}^{N} m_i \vartheta_i = -\sum_{i=1}^{N} m_i \left(\mathbf{v}_i + \overline{\boldsymbol{u}}\right) \cdot d\boldsymbol{S}_i$$

 $\overline{\boldsymbol{u}} \cdot d\boldsymbol{S}_i = \frac{d\boldsymbol{r}}{dt} \cdot \frac{d\vartheta_i}{d\boldsymbol{r}}$ 

$$\vartheta_i = \vartheta_i(\boldsymbol{r}_i(t), \boldsymbol{r}(t))$$

 $\vartheta_i = \vartheta_i(\boldsymbol{r}_i(t), \boldsymbol{r})$ 

$$\frac{d}{dt} \int_{V} \rho dV = \oint_{S} \rho \left( \boldsymbol{u} - \overline{\boldsymbol{u}} \right) \cdot d\boldsymbol{S}$$

$$\oint_{S} \rho \boldsymbol{u} \cdot d\boldsymbol{S} - \oint_{S} \rho \overline{\boldsymbol{u}} \cdot d\boldsymbol{S} = 0$$

# **Continuum Analytical Couette Flow**

![](_page_56_Figure_1.jpeg)

Where,  $\lambda_n = \left(\frac{n\pi}{L}\right)^2$  and  $u_n(t) = \frac{2U_0(-1)^n}{n\pi} \left(e^{-\frac{\lambda_n \mu t}{\rho}} - 1\right)$ **Imperial College**57
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# **Unsteady Couette Flow**

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#### Continuum Analytical

• Simplify the momentum balance (Navier-Stokes) equation

$$\frac{\partial}{\partial t}\boldsymbol{u} + \boldsymbol{\nabla} \cdot \boldsymbol{u} \boldsymbol{u} = \frac{1}{\rho} \boldsymbol{\nabla} \boldsymbol{P} + \frac{\mu}{\rho} \boldsymbol{\nabla}^2 \boldsymbol{u}$$

• Solve the 1D unsteady diffusion equation.

$$\frac{\partial u_x}{\partial t} = \frac{\mu}{\rho} \frac{\partial^2 u_x}{\partial y^2}$$

• With Boundary Conditions

$$u_x(0,t) = 0$$
$$u_x(L,t) = U_0$$
$$u_x(y,0) = 0$$

#### . Molecular Dynamics

• Fixed bottom wall, sliding top wall with both thermostatted

![](_page_57_Picture_10.jpeg)

# **Unsteady Couette Flow**

### Continuum Analytical

• Simplify the control volume momentum balance equation

$$\frac{\partial}{\partial t} \int_{V} \rho \boldsymbol{u} dV = -\oint_{S} \rho \boldsymbol{u} \boldsymbol{u} \cdot d\boldsymbol{S}$$
$$-\oint_{S} P \boldsymbol{I} \cdot d\boldsymbol{S} + \oint_{S} \boldsymbol{\sigma} \cdot d\boldsymbol{S}$$

. Simplifies for a single control volume

$$\frac{\partial}{\partial t}\int_{V}\!\!\!\!\rho u_{x}dV\!=\!\int_{S_{y}^{+}}\!\!\!\!\!\sigma_{xy}dS_{f}^{+}\!-\!\int_{S_{f}^{-}}\!\!\!\!\!\!\sigma_{xy}dS_{y}^{-}$$

• With Boundary Conditions

$$u_x(0,t) = 0$$
$$u_x(L,t) = U_0$$
$$u_x(y,0) = 0$$

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- Molecular Dynamics
  - Discrete form of the Momentum balance equation

$$\frac{d}{dt} \sum_{i=1}^{N} m_i \mathbf{v}_i \vartheta_i = -\oint_S \rho \boldsymbol{u} \boldsymbol{u} \cdot d\boldsymbol{S}$$
$$-\sum_{i=1}^{N} (\boldsymbol{v}_i - \boldsymbol{u}) (\boldsymbol{v}_i - \boldsymbol{u}) \cdot d\boldsymbol{S}_i - \sum_{i=1}^{N} \sum_{j \neq i}^{N} \varsigma_{ij} \cdot d\boldsymbol{S}_{ij}$$

• Simplifies for a single control volume

$$\frac{d}{dt}\sum_{i=1}^{N}m_i\mathbf{v}_i\vartheta_i = \sum_{i,j}^{N}f_{xij}dS_{yij}^+ - \sum_{i,j}^{N}f_{xij}dS_{yij}^-$$

• Fixed bottom wall, sliding top wall with both thermostatted

![](_page_58_Figure_15.jpeg)

## **Unsteady Couette Flow**

### . Simulation setup

- . Starting Couette flow
- · Wall thermostat: Nosé-Hoover
- Averages are computed over 1000 time steps and 8 realizations

![](_page_59_Figure_5.jpeg)

![](_page_59_Figure_6.jpeg)

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## Flow past a cylinder

. Use of the momentum conservation of the control volume to determine the drag coefficient

![](_page_60_Figure_2.jpeg)

. Drag over a Carbon Nano-tube can be determined

![](_page_60_Figure_4.jpeg)

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